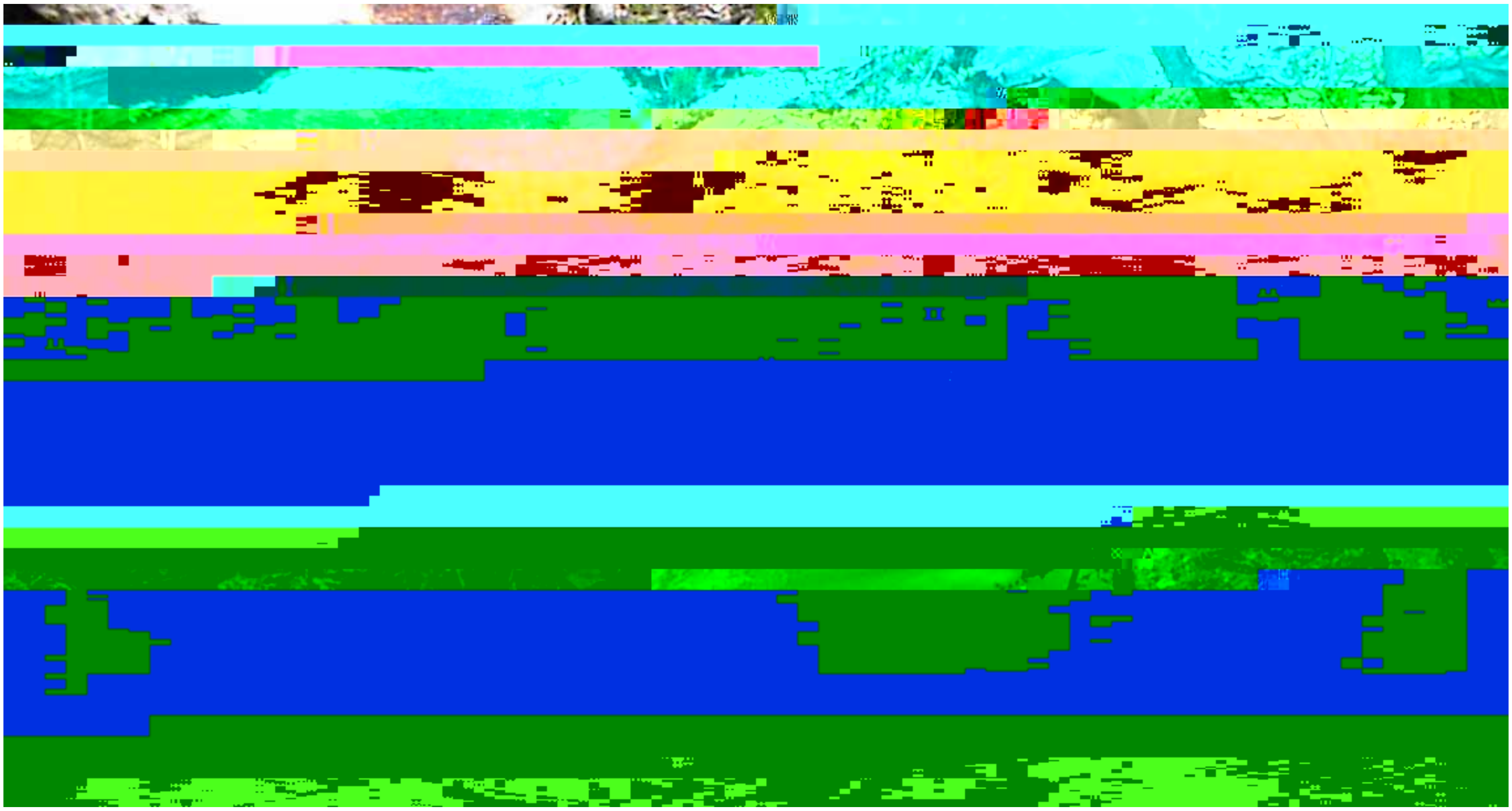


by the
Center for

and
Robert Pitt
University of Alabama

October 2004

TECHNICAL APPENDICES

ILLICIT DISCHARGE DETECTION AND ELIMINATION: A GUIDANCE MANUAL FOR PROGRAM DEVELOPMENT AND TECHNICAL ASSESSMENTS

by

Edward Brown and Deb Caraco
Center for Watershed Protection
Ellicott City, Maryland 21043

and

Robert Pitt
University of Alabama
Tuscaloosa, Alabama 35487

EPA Cooperative Agreement
X-82907801-0

Project Officer

Bryan Rittenhouse
Water Permits Division
Office of Water and Wastewater
U.S. Environmental Protection Agency
Washington, D.C.

October 2004

Table of Contents

APPENDIX A

GENERATING SITES, STORM WATER REGULATORY STATUS, AND DISCHARGE POTENTIAL

The information presented in this Appendix refers to the Standard Industrial Classification (SIC) system. This system has historically been used to classify industries and other businesses for census, tax, permit and other purposes. It should be noted that, more recently, federal agencies, including EPA, have adopted the North American Industry Classification System (NAICS, pronounced “Nakes”) as the industry classification system. For more information on the NAICS and how it correlates with SIC, visit http://www.census.gov/epcd/www/naics.html.

Overview

Identification of land uses that may impact water quality in local streams can be a difficult and time-consuming task. Research suggests that program managers might wish to preferentially investigate certain land uses when looking for the sources of possible pollutant loads. These land uses are all considered to be generating sites where routine operations can produce higher levels of storm water pollutants, and/or present a higher potential risk for spills, leaks or illicit discharges. There are two basic types of generating sites: *regulated hotspots* that are known sources of pollution and are subject to federal or state regulations, and *unregulated hotspots* which are operations suspected to be potential pollution sources, but which are not currently regulated.

Identifying Potential Generating Sites

The number and type of generating sites present in a subwatershed may vary greatly, and currently there is no public database hazardous waste), and sanitation or wastewater treatment agencies.

Phase 3. Permit Review

The final source for information is through a review of local permits. Most permit databases have SIC codes as one of the fields. These codes can be matched against the SIC codes in Table A.1 that list common generating sites under major land use headings. If a local permit database does not exist, it may be worthwhile to simply get the local phone book and do a quick look for businesses that are similar to those listed in Table A.1.

Compiling the findings from the various databases will provide an initial list of potential generating sites for future investigation. However, research has found that most of these databases can miss many of the industries that are subject to regulation (Duke *et al.*, 1999; Duke and Shaver, 1999), and further identification may be necessary. Field investigations using techniques such as the Unified Subwatershed and Site Reconnaissance (Wright *et al.*, 2004) can assist in identifying many of these generating sites that should likely be regulated by communities.

Reference Tables

This appendix is designed to assist in identifying the land uses and associated mm9 Tc0.0009 Tcw 12 y. and a *id.dr; Temr1*

Table A.1: Common Generating Sites and their Pollution Potential

	Illicit Discharge Potential*

Table A.1: Common Generating Sites and their Pollution Potential

Land Use Generating Site Description	Associated SIC Code(s)	Regulated Storm Water Hotspot	Unregulated Storm Water Hotspot	Illicit Discharge Potential*
				Direct

Table A.2: SIC and Activity Codes for EPA’s Multi-Sector General Permit for Industrial Activity
Sector A. Timber Products

Table A.2: SIC and Activity Codes for EPA’s Multi-Sector General Permit for Industrial Activity	
Sector F. Primary Metals	
3312–3317	Steel Works, Blast Furnaces, and Rolling and Finishing Mills
3321–3325	Iron and Steel Foundries
3331–3339	Primary Smelting and Refining of Nonferrous Metals
3341	Secondary Smelting and Refining of Nonferrous Metals
3351–3357	Rolling, Drawing, and Extruding of Nonferrous Metals
3363–3369	Nonferrous Foundries (Castings)
3398, 3399	Miscellaneous Primary Metal Products
Sector G. Metal Mining (Ore Mining and Dressing)	
1011	Iron Ores
1021	Copper Ores
1031	Lead and Zinc Ores
1041, 1044	Gold and Silver Oj10.02 0 0 10.02 310.24779 s9ilver Oj10.02 0.j10.02 0 0 1ng (O
1061	
1081	
1094, 1099	

Table A.2: SIC and Activity Codes for EPA’s Multi-Sector General Permit for Industrial Activity	
Sector T. Treatment Works	
TW	Treatment Works
Sector U. Food and Kindred Products	
2011–2015	
2021–2026	
2032	
2041–2048	
2051–2053	
2061–2068	
2074–2079	
2082–2087	
2091–2099	
2111–214	

References

Duke, D., K. Patel, and B. Masek. 1999. "Widespread Failure to Comply with U.S. Storm Water Regulations for Industry-Part I: Publicly Available Data to Estimate Number of Potentially Regulated Facilities." in *Environmental Engineering Science.* Vol. 16(4).

Duke, D. and K. Shaver. 1999. "Widespread Failure to Comply with U.S. Storm Water Regulations for Industry-Part II: Facility-Level Evaluations to Estimate Number of Regulated Facilities." in *Environmental Engineering Science.* Vol. 16(4).

Pitt, R. 2001. *Methods for Detection of Inappropriate Discharges to Storm Drainage Systems: Background Literature and Summary of Findings*. Tuscaloosa, AL.

Wright, T., C. Swann, K. Cappiella, and T. Schueler. 2004. *Unified Subwatershed and Site Reconnaissance: A User's Manual*. Center for Watershed Protection. Ellicott City, MD.

Appendix B: Model Illicit Discharge and Connection Ordinance

Illegal Discharge. Any direct or indirect non-storm water discharge to the storm drain system, except as exempted in Section 8 of this ordinance.

Illicit Connections. An illicit connection is defined as either of the following:

- Any drain or conveyance, whether on the surface or subsurface that allows an illegal discharge to enter the storm drain system including but not limited to any conveyances that allow any non-storm water discharge including sewage, process wastewater, and wash water to enter the storm drain system and any connections to the storm drain system from indoor drains and sinks, regardless of whether said drain or connection had been previously allowed, permitted, or approved by an authorized enforcement agency or,
- Any drain or conveyance connected from a commercial or industrial land use to the storm drain system that has not been documented in plans, maps, or equivalent records and approved by an authorized enforcement agency.

Industrial Activity. Activities subject to NPDES Industrial Storm Water Permits as defined in 40 CFR, Section 122.26 (b)(14).

Municipal Separate Storm Sewer System (MS4)

Storm Water. Any surface flow, runoff, and drainage consisting entirely of water from any form of natural precipitation, and resulting from such precipitation.

Storm Water Management Plan. A document which describes the Best Management Practices and activities to be implemented by a person or business to identify sources of pollution or contamination at a site and the actions to eliminate or reduce pollutant discharges to Storm Water, Storm Water Conveyance Systems, and/or

SECTION 8. DISCHARGE PROHIBITIONS.

8.1. Prohibition of Illegal Discharges.

No person shall throw, drain, or otherwise discharge, cause, or allow others under its control to throw, drain, or otherwise discharge into the MS4 any pollutants or waters containing any pollutants, other than storm water.

The commencement, conduct or continuance of any illegal discharge to the storm drain system is prohibited except as described as follows:

(1) The following discharges are exempt from discharge prohibitions established by this ordinance: water line flushing, landscape irrigation, diverted stream flows, rising ground waters, uncontaminated ground water infiltration, uncontaminated pumped ground water, discharges from potable water sources, foundation drains, air conditioning condensation, irrigation water, springs, water from crawl space pumps, footing drains, lawn watering, individual residential car washing, flows from riparian habitats and wetlands, dechlorinated swimming pool discharges, and street wash water.

(2) Discharges or flow from firefighting, and other discharges specified in writing by the **[authorized enforcement agency]** as being necessary to protect public health and safety.

(3) Discharges associated with dye testing, however this activity requires a verbal notification to the **[authorized enforcement agency]**

enforcement agency] requiring that such locating be completed. Such notice will specify a reasonable time period within which the location of the drain or conveyance is to be determined, that the drain or conveyance be identified as storm sewer, sanitary sewer or other, and that the outfall location or point of connection to the storm sewer system, sanitary sewer system or other discharge point be identified. Results of these investigations are to be documented and provided to the **[authorized enforcement agency]**.

SECTION 9. WATERCOURSE PROTECTION.

Every person owning property through which a watercourse passes, or such person's lessee, shall keep and maintain that part of the watercourse within the property free of trash, debris, excessive vegetation, and other obstacles that would pollute, contaminate, or significantly retard the flow of water through the watercourse. In addition, the owner or lessee shall maintain existing privately owned structures within or adjacent to a watercourse, so that such structures will not become a hazard to the use, function, or physical integrity of the watercourse.

SECTION 10. INDUSTRIAL OR CONSTRUCTION ACTIVITY DISCHARGES.

10.1. Submission of NOI to [jurisdiction].

(1) Any person subject to an industrial or construction activity NPDES storm water discharge permit shall comply with all provisions of such permit. Proof of compliance with said permit may be required in a form acceptable to the **[authorized enforcement agency]** prior to the allowing of discharges to the MS4.

(2) The operator of a facility, including construction sites, required to have an NPDES permit to discharge storm water associated with industrial activity shall submit a copy of the Notice of Intent (NOI) to the **[authorized enforcement agency]** at the same time the operator submits the original Notice of Intent to the EPA as applicable.

(3) The copy of the Notice of Intent may be delivered to the

3authopr7zedrpnforc agency]

SECTION 12. REQUIREMENT TO PREVENT, CONTROL, AND REDUCE STORM WATER POLLUTANTS BY THE USE OF BEST MANAGEMENT PRACTICES.

[Authorized enforcement agency] will adopt requirements identifying Best Management Practices for any activity, operation, or facility which may cause or contribute to pollution or contamination of storm water, the storm drain system, or waters of the United States. The owner or operator of such activity, operation, or facility shall provide, at their own expense, reasonable protection from accidental discharge of prohibited materials or other wastes into the municipal storm drain system or watercourses through the use of these stru without giving prior notice, to take any and all measures necessary to abate the violation and/or restore the property. The **[authorized enforcement agency]** is authorized to seek costs of the abatement as outlined in Section 17.

14.2. Warning Notice.

When the **[authorized enforcement agency]** finds that any person has violated, or continues to violate, any provision of this ordinance, or any order issued hereunder, the **[authorized enforcement agency]** may serve upon that person a written Warning Notice, specifying the particular violation believed to have occurred and requesting the discharger to immediately investigate the matter and to seek a resolution whereby any offending discharge will cease. Investigation and/or resolution of the matter in response to the Warning Notice in no way relieves the alleged violator of liability for any violations occurring before or after receipt of the Warning Notice. Nothing in this subsection shall limit the authority of the **[authorized enforcement agency]** to take any action, including emergency action or any other enforcement action, without first issuing a Warning Notice.

14.3. Notice of Violation.

Whenever the **[authorized enforcement agency]** finds that a person has violated a prohibition or failed to meet a requirement of this ordinance, the **[authorized enforcement agency]** may order compliance by written notice of violation to the responsible person.

The Notice of Violation shall contain:

(1) The name and address of the alleged violator;

(2) The address when available or a description of the building, structure or land upon which

restoration of any affected property

(5) Payment of a fine to cover administrative and remediation costs; and

(6) The implementation of source control or treatment BMPs.

14.5. Compensatory Action.

In lieu of enforcement proceedings, penalties, and remedies authorized by this ordinance, the **[authorized enforcement agency]** may impose upon a violator alternative compensatory actions, such as storm drain stenciling, attendance at compliance workshops, creek cleanup, etc.

14.6. Suspension Of MS4 Access.

comply with a suspension order issued in an emergency, the **[authorized enforcement agency]** may take such steps as deemed necessary to prevent or minimize damage to the MS4 or waters of the United States, or to minimize danger to persons.

14.6.3. Suspension due to the Detection of Illicit Discharge

Any person discharging to the MS4 in violation of this ordinance may have their MS4 access terminated if such termination would abate or reduce an illicit discharge. The **[authorized enforcement agency]** will notify a violator of the proposed termination of its MS4 access. The violator may petition the **[authorized enforcement agency]** for a reconsideration and hearing.

premises to refuse to allow the government agency or designated contractor to enter upon the premises for the purposes set forth above.

SECTION 17. COST OF ABATEMENT OF THE VIOLATION.

Within **[___]** days after abatement of the violation, the owner of the property will be notified of the cost of abatement, including administrative costs. The property owner may file a written protest objecting to the amount of the assessment within **[___]** days. If the amount due is not paid within a timely manner as determined by the decision of the municipal authority or by the expiration of the time in which to file an appeal, the charges shall become a special assessment against the property and shall constitute a lien on the property for the amount of the assessment.

Any person violating any of the provisions of this article shall become liable to the **[jurisdiction]**

APPENDIX C

SIX STEPS TO ESTABLISHING A HOTLINE AND REPORTING AND TRACKING SYSTEM

Introduction

A complaint hotline is a dedicated phone number or website where citizens can easily report illicit discharge and pollution concerns. A prompt investigation of each complaint by trained inspectors should always follow a reported incident, usually within 24 hours. Many Phase I communities utilize hotlines to track down intermittent and transitory discharges, and regard them as one of their most effective tools to isolate illicit discharges (CWP, 2002).

This appendix describes the six steps needed to establish a hotline to report and track illicit discharges.

Step 1. Define the scope

The community must first determine its need for an IDDE complaint hotline and should not establish one simply because it does not currently exist. An IDDE hotline may be appropriate for a community for the following reasons:

- The municipality already receives a high volume of complaint calls associated with illicit discharges. Without a designated number, complaints may be received by several different departments, which can lead to inconsistent handling of concerns. If a community is unsure of the number of complaints received across the municipality, it may want to quickly survey departments likely to receive calls. A hotline can help promote stakeholder reporting of incidents and make the reporting process more efficient rather than relying on calls making it to the correct office.
- A community hotline exists that cannot be altered to accommodate the needs of the IDDE program. Situations that would make two hotlines incompatible include: significantly different concerns (e.g., IDDE vs. emergency services); varying jurisdictional limits (e.g., regional vs. city only); and funding restrictions (e.g., hotline is developed with a grant that prevents it from overlapping with other programs).
- Related municipal programs exist that would benefit from the establishment of a hotline, such as erosion and sediment control or storm water management programs. Combining similar services can lead to a significant savings in cost and time.

Communities that have no pressing need for a hotline may still choose to institute a department phone number or email address to field complaints and incident reports during normal business hours, or a website that provides guidance on how to report potential illicit discharges.

Once a community has decided to implement a hotline, the scope of the IDDE hotline should be defined, including the intent and extent of the program. The intent of the hotline may be to process the incident/complaint, and investigate and enforce violations, or to take a more educational approach that also provides information and guidance. It is recommended that communities initiating a hotline for the first time limit the scope to the former intent.

The extent of the hotline refers to the geographic area of coverage as well as the types of incidents that fall under the responsibility of the responding agency or department. Often hotlines are restricted to one specific jurisdiction to minimize complications with investigating and enforcing violations across jurisdictional lines. Significant coordination and planning are required if the hotline is intended to serve a region or watershed with several jurisdictions. Similar coordination efforts are necessary if a wide range of incidents is handled by the hotline that require multiple agencies or departments to respond. It is important for communities to predetermine what agency or department is best suited and trained to respond to specific incident reports, and for all hotline operators to be well trained and knowledgeable about these distinctions.

Step 2. Create a tracking and reporting system

The next step to establishing an IDDE hotline is to create a tracking and reporting system. The two key features that should be considered are the methods of reporting and methods of responding.

At a minimum, the reporting method should include a telephone call-in system and may also include a website. The phone number and/or internet address should be easy to remember and toll-free if any areas under the jurisdiction of the IDDE program are long-distance from the reporting office. The reporting method should be available 24 hours a day, seven days a week. This around the clock process encourages stakeholders to call as soon as a problem is identified.

Providing an option for anonymous reporting also encourages calls because it can be done without fear of retribution from neighbors, employers, or others. In most cases this is achieved by providing an “Incident ID” that may also be used to allow the caller to track the investigation and know that their concerns are being addressed, as well as build in accountability within the department to respond to hotline callers.

The level of detail collected during an incident report will vary depending on system design and complaint responder training. Many hotlines collect only basic information, however, more detailed information will help prioritize investigations and take advantage of a database system to expand reporting options. A sample Illicit Discharge Incident Tracking Sheet is provided at the end of this Appendix to help facilitate this process. The sheet is intended for use with a phone reporting system, and is designed so that the responder can prompt the caller through each section. This sheet may be modified into a simple, multiple-choice questionnaire if reporting is done through a website or email. The basic information collected during an incident report is described below.

- *Incident ID* - Each incident should receive a unique identification code to ensure accurate tracking and public feedback.

- *Reporter Information* - Reporter contact and visible outfall numbering allows reports to be precisely located. In the absence of outfall IDs, callers should be encouraged to provide the nearest street/intersection information and any general descriptions that tie the site to a nearby landmark or major land use (e.g., shopping center, school, etc.), as well as indicate whether the incident site is located in the stream corridor or in an upland area. Other options are to include blank space for narrative descriptions or for the response team to meet the caller at a nearby known location if the caller cannot provide sufficient locational information.

- *Problem Type* - Providing a list of likely problems and problem descriptions can help to readily identify the potential source. The problem types will likely fall into the following five categories: unnatural stream conditions, sewage, wash water, oil/solvents, and industrial wastes. “Other” should also be included, as exceptions will occur. By identifying a suspected origin, the field team may have a head start on the investigation and suspected repeat offenders can be screened through trend analysis.

- *Problem Indicator Description* - A description of the discharge odor and color, and type of floatables present permits investigators to know what they are looking for and start preparing for how to handle the situation.

- *Investigation Notes* - To properly track and report suspected illicit discharges, the investigation needs to be documented. Key information to record for the initial and follow-up investigation (if applicable) include: date, time, step taken to respond to incident report (not all require follow up), investigators, length of time spent for investigation, corrective actions taken, date case closed, and any other pertinent information.

Due to the intermittent nature of illicit discharges, a 24-hour investigation response can increase the likelihood of identifying and eliminating problems. While some problems require more immediate attention than others, investigators should always respond as soon as appropriate. Calls should be screened by a “live” person so only the most urgent calls are passed through a pager system in order to minimize the pressure that 24-hour response places on investigators at odd hours. The complaint questions should be detailed enough to help support this basic prioritization.

Some communities may determine that 24-hour response is cost prohibitive, and that non-emergency response will only occur during normal working hours (e.g., 8AM - 5PM). In these situations, it is essential that trained so that he/she understands the significance of the information being collected and can go beyond the "check boxes" when necessary to answer the reporter questions, as well as guide the caller through the data collection process. This ensures that the incident is handled correctly, and that the caller feels that the concern is in good hands.

An initial screening of the potential illicit source by the responder can be useful. Table C1 provides a list of descriptions of common illicit discharges called in and the likely source or situation.

Inter- and intra-department training should focus on the importance of IDDE, the complaint hotline investigation and tracking process, and the expected responsibilities of each involved entity. Such training can greatly increase watershed wide awareness of illicit discharge problems and is essential to developing good working relationships with other departments.

Step 4. Advertise

Public relations are an important aspect of a pollution hotline. Many municipalities have noted that there is always a peak in incident reporting following an advertising campaign. Advertising the hotline phone number or web address several times a year keeps the message fresh in public minds. Effective methods include magnets, stickers, phone book advertisements, flyers, bill inserts, displays, fair booths and newspaper articles.

Advertising, including publicizing success stories about the hotline serves several purposes. First it highlights the responsiveness of the program to the general public. Second, it serves as a means to further promote the hotline. Third, it builds public support for the program and fosters public stewardship. Success stories can be published through newspaper articles, TV broadcasts or other highly visible means of advertising. The stories will build general awareness of illicit discharge issues and promote greater public stewardship and accountability by both those reporting the problems and potential violators.

Table C1: Types of Potential IDDE Hotline Complaints

Typical Call-in Indicators	Likely Source
Sewage smell, or floatables from storm drain outfall during dry weather flow	Storm and sanitary sewer cross-connection

Step 5. Respond to complaints

Hotline customer service staff should provide friendly and knowledgeable service to callers that might include an overview of the investigation process, how long a response should take, and an incident tracking ID so the caller can follow-up on the complaint. Hotline staff should arrange to send an investigator out to the incident site as soon as possible.

Investigators should respond to complaints in a timely manner, and provide the necessary feedback to the database system. The type of complaint will dictate the necessary response, as well as the timing of the response (e.g., a failing septic system may not be as high a priority as a sanitary sewer overflow). Information submitted to the reporting database might include: timTh9 foos0.001ni

Table C2: Cost to Create and Maintain a Successful IDDE Hotline

Steps	

References

Center for Watershed Protection (CWP).
1998.

Illicit Discharge Hotline Incident Tracking Sheet

Incident ID:

Responder Information

Investigation Notes

Initial investigation date:

Investigators:

APPENDIX D

OUTFALL RECONNAISSANCE INVENTORY FIELD SHEET

OUTFALL RECONNAISSANCE INVENTORY/ SAMPLE COLLECTION FIELD SHEET

Section 1: Background Data

Subwatershed:		Outfall ID:	
Today's date:		Time (Military):	
Investigators:		Form completed by:	
Temperature (°F):	Rainfall (in.): Last 24 hours: Last 48 hours:		
Latitude:	Longitude:	GPS Unit:	GPS LMK #:
Camera:		Photo #s:	

Land Use in Drainage Area (Check all that apply):

☐ Industrial

☐ Ultra-Urban Residential

☐ Suburban Residential

☐ Commercial

☐ Open Space

☐ Institutional

Other: ____

Known Industries: ____

Notes (e.g., origin of outfall, if known):

Section 2: Outfall Description

LOCATION	MATERIAL		SHAPE		DIMENSIONS (IN.)	SUBMERGED
☐ **Closed Pipe**	☐ RCP ☐ PVC ☐ Steel ☐ Other: __________	☐ CMP ☐ HDPE	☐ Circular ☐ Eliptical ☐ Box ☐ Other: _____	☐ Single ☐ Double ☐ Triple ☐ Other: _____	Diameter/Dimensions: ______________________	In Water: ☐ No ☐ Partially ☐ Fully With Sediment: ☐ No ☐ Partially ☐ Fully
☐ **Open drainage**	☐ Concrete ☐ Earthen ☐ rip-rap ☐ Other: _____		☐ Trapezoid ☐ Parabolic ☐ Other: _____		Depth: _____ Top Width: _____ Bottom Width: _____	▨
☐ **In-Stream**	**(applicable when collecting samples)**					
Flow Present?	☐ Yes	☐ No	***If No, Skip to Section 5***			
Flow Description (If present)	☐ Trickle	☐ Moderate	☐ Substantial			

Section 3: Quantitative Characterization

FIELD DATA FOR FLOWING OUTFALLS		
PARAMETER	RESULT	

Outfall Reconnaissance Inventory Field Sheet

Section 4: Physical Indicators for Flowing Outfalls Only

Are Any Physical Indicators Present in the flow? ☐ Yes ☐

APPENDIX E

FLOW TYPE DATA FROM TUSCALOOSA AND BIRMINGHAM, AL

Table E1.1: Tap Water Reference ("Library") Samples

Sample number	Sampling Location	Date	pH	Spec. cond. (µS/cm)	Temp. (°F)	Turb. (NTU)	Color (APHA Platinum Cobalt Units)	F (mg/L)	Hard. (mg/L $CaCO_3$)	Detergent (mg/L as MBAS)

Table E1.1: Tap Water Reference (“Library”) Samples, CONT.

Sample number	Sampling Location	Date	K (mg/L)	NH_3 (mg/L as N)	NH_3/K (ratio)	B (mg/L)	Total Coliforms (MPN/100 mL)	*E. coli* (MPN/100 mL)	Enterococci (MPN/100 mL)
1	B.B.Commer Hall	5/17/2002	1	<LD	N/A	N/A	N/A	N/A	N/A
2	Rose Towers	5/17/2002	1	<LD	N/A	N/A	N/A	N/A	N/A
3	H.C.Commer Hall	5/17/2002	1	<LD	N/A	N/A	N/A	N/A	N/A
4	Rec Centre	5/17/2002	1	<LD	N/A	N/A	N/A	N/A	N/A
5	Coleman Coliseum	5/17/2002	1	<LD	N/A	N/A	N/A	N/A	N/A
6	Mib (UA)	5/29/2003	2	0.01	0.005	0.19	1	<1	<1
7	Alex Appt.	5/30/2003	2	<LD	N/A	0.1	<1	<1	<1
8	Georgas Library (UA)	6/3/2003	1	<LD	N/A	0.12	<1	<1	<1
9	Rodgers Library	6/8/2003	1	<LD	N/A	0.04	21.6	<1	<1
10	Alexander Property Appt.	6/8/2003	1	0.07	0.07	0.14	<1	<1	<1
11	Pslidea Court Appt.	6/8/2003	2	0.07	0.035	0.27	<1	<1	<1
12	University Plaza Appt.	6/8/2003	2	0.07	0.035	0.11	<1	<1	<1
Mean			1.3	<0.055	0.036	0.14	<11	<1	<1
Standard Deviation			0.49	0.03	0.026	0.07	15	-	-
COV			0.37	0.55	0.73	0.53	1.3	-	-
Anderson Darling Probability Test Value (Normal)			**3.809**	**3.199**	**2.539**	**1.663**	4.103	-	-
Anderson Darling Probability Test Value (Log-normal)			3.809	3.199	2.703	1.685	4.103	-	-

Data provided by Robert Pitt, University of Alabama

Table E1.2: Spring Water Reference ("Library") Samples

Table E1.3: Car Wash Reference ("Library") Samples

Sample number	Sampling Location	Date	pH

Table E1.4: Laundry Reference ("Library") Samples

Sample number	Sampling Location	Date	pH	Spec. cond. (µS/cm)	Temp. (°F)	Turb. (NTU)	Color (APHA Platinum Cobalt Units)	F (mg/L)	Hard. (mg/L $CaCO_3$)	Detergent

0.8 g148.2 594.20 594.24 9.24 44.52 refBT0

Table E1.5: Sewage (Dry Weather) Reference ("Library") Samples

Sample number	Sampling Location	Date	pH	Spec. cond. (µS/cm)	Temp. (°

Table E1.8: Industrial (Cintas) Reference ("Library") Samples

Sample number

Table E1.9: Irrigation Reference ("Library") Samples

Sample number	Sampling Location	Date	pH	Spec. cond. (μS/cm)	Temp. (°F)	Turb. (NTU)	Color (APHA Platinum Cobalt Units)	F (mg/L)	Hard. (mg/L $CaCO_3$)	Detergent (mg/L as MBAS)	Fluorescence (raw signal strength)	Fluorescence (mg/L as "Tide")
1	Ferguson Parking (UA) - Run over concrete	5/16/2003	7.91	200	N/A	16.2	0	0.69	62	0	21226	49
2	B.B. Commer (UA) - Run over concrete	5/18/2003	7.38		N/A	4.03	10	0.68	60	0	13915	32
3	Art Building (UA) - Taken at a little puddle, NO concrete	5/16/2003	7.46	200	N/A	64.6	0	0.76	55	0	40040	92
4												

Table E1.9: Irrigation Reference (“Library”) Samples, CONT.

Sample number	Sampling Location	Date	K (mg/L)	NH_3 (mg/L as N)	NH_3/K (ratio)	B (mg/L)	Total Coliforms (MPN/100 mL)	*E. coli* (MPN/100 mL)	Enterococci (MPN/100 mL)
1	Ferguson Parking (UA) - Run over concrete	5/16/2003	2	<LD	N/A	0.14	>2419.2	27.8	>2419.2
2	B.B. Commer (UA) - Run over concrete	5/18/2003	9	1.0	0.111	0.20	>2419.2	8.3	2
3	Art Building (UA) - Taken at a little puddle, NO concrete	5/16/2003	5	0.08	0.016	0.25	>2419.2	>2419.2	>2419.2
4	MIB (UA) - Run over concrete	5/19/2003	3	0.21	0.07	0.13	>2419.2	>2419.2	>2419.2
5	MIB (UA) - Run over concrete	5/30/2003	2	3.5	1.75	0.2	>2419.2	31.8	>2419.2
6	Art Building (UA) -Taken at a little puddle, NO concrete	5/30/2003	4	0.5	0.125	0.36	>2419.2	>2419.2	287.7
7	Quad(UA) - Taken at a little puddle, NO concrete	5/30/2003	5	1	0.2	0.5	>2419.2	>2419.2	>2419.2
8	MIB (UA) - Run over concrete	6/5/2003	9	4.5	0.5	0.22	>2419.2	>2419.2	>2419.2
9	MIB (UA) - Taken at a little puddle, NO concrete	6/5/2003	8	0.5	0.06	0.14	>2419.2	>2419.2	>2419.2
10	Bevil (UA) - Taken at a little puddle, NO concrete	6/5/2003	4	1	0.25	0.23	>2419.2	1299.7	>2419.2
11	MIB (UA) - Run over concrete	6/9/2003	7	0.5	0.07	0.25	>4838.4	>4838.4	>4838.4
12	MIB (UA) - Taken at a little puddle, NO concrete	6/9/2003	10	1	0.1	0.35	>4838.4	>4838.4	>4838.4
Mean			5.6	1.25	0.29	0.24	>2419.2	>2419.2	>2419.2
Standard Deviation			2.8	1.41	0.50	0.10	-	-	-
COV			0.50	1.12	1.69	0.43	-	-	-
Anderson Darling Probability Test Value (Normal)			**1.144**	2.471	3.343	1.366	-	-	-
Anderson Darling Probability Test Value (Log-normal)			1.146						

Appendix E-2. Data Tables for Birmingham

Table E2.2: Shallow Ground Water Samples

Sample #	Conductivity (µS/cm)	Fluoride (mg/L)	Hardness (mg/L) (as $CaCO_3$)	Detergent (mg/L)	Fluoresc. (% scale)	Potassium (mg/L)	Ammonia (mg/L)	pH	Color (units)	Chlorine (mg/L)	Toxicity (I25) (% reduc.)	Copper (mg/L)	Phenols (mg/L)
1	5	0.08	5	0	7	NA	NA	NA	5	0.04	0	0.01	0
2	5	0.03	22	0	12	NA	NA	NA	20	0.00	0	0.01	0
3	32	0.14	18	0	160	NA	NA	7.8	35	0.08	0	0.00	0
4	128	0.07	41	0	34	1.70	0.38	6.2	0	0.02	0	0.00	0
5	119	0.05	38	0	22	2.15	0.89	5.4	0	0.00	0	0.00	0

Table E2.3: Samples from Irrigation of Landscaped Areas

Sample #	Conductivity (μS/cm)	Fluoride (mg/L)	Hardness (mg/L) (as $CaCO_3$)	Detergent (mg/L)	Fluoresc. (% scale)	Potassium (mg/L)	Ammonia (mg/L)	pH	Color (units)	Chlorine (mg/L)

Table E2.4: Residential/Commercial Sanitary Sewage Samples

Sample #	Collection Date	Collection Time	Conductivity (µS/cm)	Fluoride (mg/L)	Hardness (mg/L) (as $CaCO_3$)	Detergent (mg/L)	Fluoresc. (% scale)	Potassium (mg/L)
1	1-Aug	10 p.m.	265	0.90	149	0.96	240	5.25
2	2-Aug	12 a.m.	320	0.72	161	3.80	200	4.79
3	2-Aiug	2 a.m.	360	0.46	172	0.58	170	3.44
4	2-Aug	4 a.m.	350	0.58	181	0.54	155	3.09
5	2-Aug	6 a.m.	410	0.74	167	0.54	205	4.51
6	2-Aug	8 a.m.	435	0.87	154	0.99	265	5.88
7	2-Aug	10 a.m.	410	1.08	150	0.48	265	5.99

Table E2.5: Residential Septic Tank Discharge Samples

Sample #	Conductivity (µS/cm)	Fluoride (mg/L)	Hardness (mg/L) (as $CaCO_3$)	Detergent (mg/L)

Table E2.7: Commercial Laundry Samples

Table E2.8: Radiator Waste Samples

Sample #	Conductivity (µS/cm)	Fluoride (mg/L)	Hardness (mg/L) (as $CaCO_3$)	Detergent (mg/L)	Fluoresc. (% scale)	Potassium (mg/L)	Ammonia (mg/L)	pH

Table E2.9: Plating Bath Waste Samples

Sample #	Conductivity (µS/cm)	Fluoride (mg/L)	Hardness (mg/L) (as $CaCO_3$)	Detergent (mg/L)	Fluoresc. (% scale)

APPENDIX F

ANALYTICAL PROCEDURES FOR OUTFALL MONITORING

APPENDIX F1: INDICATOR PARAMETER OVERVIEW

Ammonia

Ammonia is a good indicator of sewage, since its concentration is much higher there than in groundwater or tap water. High ammonia concentrations may also indicate liquid wastes from some industrial sites. Ammonia is relatively simple and safe to analyze. Some challenges include the tendency for ammonia to volatilize (i.e., turn into a gas and become non-conservative) and its potential generation from non-human sources, such as pets or wildlife.

Boron

Boron is an element present in the compound borax, which is often found in detergent and soap formulations. Consequently, boron is a good potential indicator for both laundry wash water and sewage. Preliminary research from Alabama supports this contention, particularly when it is combined with other detergent indicators, such as surfactants (Pitt, IDDE Project Support Material). Boron may not be a useful indicator everywhere in the country since it may be found at elevated levels in groundwater in some regions and is a common ingredient in water softeners products. Program mangers should collect data on boron concentrations in local tap water and groundwater sources to confirm whether it will be an effective indicator of illicit discharges.

Chlorine

Chlorine is used throughout the country to disinfect tap water, except where private wells provide the water supply. Chlorine

Fluoride

Fluoride is added to drinking water supplies in most communities to improve dental health, and normally found at a concentration of two parts per million in tapwater. Consequently, fluoride is an excellent conservative indicator of tap water discharges or leaks from water supply pipes that end up in the storm drain. Fluoride is obviously not a good indicator in communities that do not fluoridate drinking water, or where individual wells provide drinking water. One key constraint is that the reagent used in the recommended analytical method for fluoride is considered a hazardous waste, and must be disposed of properly.

Hardness

Hardness measures the positive ions dissolved in water and primarily include magnesium and calcium in natural waters, but are sometimes influenced by other metals. Field testing by Pitt (IDDE Project Support Material) suggests that hardness has limited value as an indicator parameter, except when values are extremely high or low (which may signal the presence of some liquid wastes). Hardness may be applicable in communities where hardness levels are elevated in groundwater due to karst or limestone terrain. In these regions, hardness can help distinguish natural groundwater flows present in outfalls from tap water and other flow types.

pH

Most discharge flow types are neutral, having a pH value around 7, although groundwater concentrations can be somewhat variable. pH is a reasonably good indicator for liquid wastes from industries, which can have very high or low pH (ranging from 3 to 12). The pH of residential wash water tends to be rather basic (pH of 8 or 9). The pH of a discharge is very simple to monitor in the field with low cost test strips or probes. Although pH data is often not conclusive by itself, it can identify problem outfalls that merit follow-up investigations using more effective indicators.

Potassium

Potassium is found at relatively high concentrations in sewage, and extremely high concentrations in many industrial process waters. Consequently, potassium can act as a good first screen for industrial wastes, and can also be used in combination with ammonia to distinguish wash waters from sanitary wastes. (See Chapter 12). Simple field probes can detect potassium at

Surfactants

Table F1: Summary of Research Indicators Used for Identifying Inappropriate Discharges into Storm Drainage

Parameter Group	Comments	Recommendation
Coprostanol and other fecal sterol compounds	Used to indicate presence of sanitary sewage	Possibly useful. Expensive analysis with GC/MSD. Not specific to human wastes or recent contamination. Most useful when analyzing particulate fractions of wastewaters or sediments.
Specific detergent compounds (LAS, fabric whiteners, and perfumes)	Used to indicate presence of sanitary sewage	Possibly useful. Expensive analyses with HPLC. A good and sensitive confirmatory method.
Pharmaceuticals (colfibric acid, aspirin, ibuprofen, steroids, illegal drugs, etc.)	Used to indicate presence of sanitary sewage	Possibly useful. Expensive analyses with HPLC. A good and sensitive confirmatory method.
Caffeine	Used to indicate presence of sanitary sewage	Not very useful. Expensive analyses with GC/MSD. Numerous false negatives, as typical analytical methods not suitably sensitive.
DNA profiling of microorganisms	Used to identify sources of microorganisms	Likely useful, but currently requires extensive background information on likely sources in drainage. Could be very useful if method can be simplified, but with less specific results.
UV absorbance at 228 nm	Used to identify presence of sanitary sewage	Possibly useful, if UV spectrophotometer available. Simple and direct analyses. Sensitive to varying levels of sanitary sewage, but may not be useful with dilute solutions. Further testing needed to investigate sensitivity in field trials.
Stable isotopes of oxygen	Used to identify major sources of water	May be useful in area having distant domestic water sources and distant groundwater recharge areas. Expensive and time consuming procedure. Can not distinguish between wastewaters if all have common source.
GC/MSD - Gas Chromatography/Mass Selective Detector HPLC - High Performance Liquid Chromatography		

Appendix F2: “Off-the Shelf” Analytical Methodologies

F2.1 AMMONIA (0 TO 0.50 MG/L NH_3-N)

Equipment/Supplies Needed

- Hach bench top or portable spectrophotometer or colorimeter (see ordering information below)
- ammonia nitrogen reagent set for 25-mL samples
- ammonia nitrogen standard solution

Procedure

Refer to Hach method 8155 for Nitrogen, Ammonia Salicylate Method (0 to 0.50 mg/L NH_3-N) for a 25mL sample. In this method, ammonia compounds combine with chlorine to form monochloramine. Monochloramine reacts with salicylate to form 5-aminosalicylate. The 5-aminosalicylate is oxidized in the presence of sodium nitroprusside catalyst to form a blue-colored compound. The blue color is masked by the yellow color from the excess reagent present to give a final green-colored solution.

Duration of Test for Each Sample

Because of the duration of this test, samples should be run in batches of about six. From start to finish, each batch of six samples takes about 25 minutes, including the time taken to clean the sample cells and reset the instrument between each batch.

Hazardous Reagents

According to good laboratory practice, the contents of each sample cell, after the analysis, should be poured into another properly-labeled container for proper disposal.

Ease of Analysis

This procedure is time-consuming and should be performed indoors.

Ordering Information

Vendor: Hach Company
PO Box 389
Loveland, CO 80539-0389
Tel: 800-227-4224
Fax: 970-669-2932
Website: www.hach.com

[Note: The direct-Nessler method may be preferred due to its faster reaction times, but Nessler reagent is toxic and corrosive. Nessler reagent, according to its MSDS, causes severe burns, is an acute and a cumulative poison, and is a teratogen. It also contains from 5 to 10% mercuric iodide. It is now recommended that the more sensitive salicylate method because of the lower concentrations experienced in this research, and because of its lower toxicity and easier disposal requirements. The salicylate method was therefore used for this project, although prior research found it to be somewhat less satisfactory than the Nessler method.]

Equipment/Supplies Needed for Ammonia Analysis		
Item (Catalog Number)	**Quantity**	**Price**
One of the colorimeters, or spectrophotometers, listed previously will be needed. Alternatively, a dedicatedcto c(thAlte6(pr 2 d)5.66m(ent 5 Tcwid. fB66.36 el)6.310.64 165s a terato28te)5.3on(b		

F2.3 COLOR (0 – 100 APHA Platinum Cobalt Units)

Equipment/Supplies needed

One Hach color test kit Model CO-1 which measures color using a color disc for comparison.

Procedure

The following procedure is described in the test kit.

Low Range

1. Place the lengthwise viewing adapter in the comparator.
2. fill one sample tube to the line underlining “Cat. 1730-00” with the sample. This will be approximately 15mL. If not using 1730-00 tubes, fill to the line founds at approximately 3 inches up from the bottom of the tube.
3. Place the tube containing the water sample into the comparator in the right-hand position.
4. Fill the other sample tube with colorless water to the line underlining “Cat. 1730-00.” Insert this tube in the left-side comparator opening.
5. Hold the comparator with the tube tops pointing to a window or light source at approximately a 45 degree angle (with the light coming in through the top of the tubes). View through the openings in the front of the comparator. When viewing, use care to not sp7al Mothrough the tounsesper0 -underliubeeRot-1.

Ease of Analysis

This procedure may be performed outside of a standard laboratory, if well ventilated. Produces hazardous chemicals.

Ordering Information

Vendor: *CHEMetrics*, Inc
4295 Catlett Rd
Calverton, VA 20138
Phone 1-800-356-3072
FAX 1-540-788-4856
Website: www.chemetrics.com

Equipment/Supplies Needed for Detergents Analysis

F2.6 *E. COLI*

Equipment/Supplies Needed

- Colilert reagent, sterile sample bottles for 100 mL samples
- Quanti-Tray 2000
- Colilert comparator predispensed in a Quanti-Tray/2000incubator
- UV light from IDEXX.

Enumeration Procedure

1. Add contents of one Colilert snap pack to a 100 mL room temperature water sample in a sterile vessel. The standard Colilert reagent is recommended when evaluating Enterococci simultaneously so the samples are both ready to read in 24 hours. If only *E. coli* are to be evaluated, then the faster Colilert-18 reagent can be used if reading the results in 18 hours instead of 24 hours is important.
2. Cap vessel and shake until dissolved.
3. Pour sample/reagent mixture into a Quanti-Tray/2000 and seal in an IDEXX Quanti-Tray Sealer.
4. Place the sealed tray in a 35±0.5^{o} C incubator for 24 hours.
5. Read results according to the Results Interpretation table below. Count the number of positive wells and refer to the MPN table provided with the Quanti-Trays to obtain a Most Probable Number.

Results Interpretation

Appearance	Result
Less yellow than the comparator	Negative for total coliforms and *E. coli*
Yellow equal to or greater than the comparator	Positive for total coliforms
Yellow and fluorescence equal to or greater than the comparator	Positive for *E. coli*

Duration of Test for Each Sample

Once the Quanti-Tray sealer is warm (10 min), it takes approximately 5 minutes per sample to label, seal and incubate the Quanti-Tray. After 24 hours, it takes 1-2 minutes to read the sample results under the UV lamp.

Hazardous Reagents

Used Quanti-Trays must be disposed of in a biohazard bag and handled by appropriate biohazard disposal facility, using similar practices as for alternative bacteria analysis methods.

Ease of Analysis

Not a difficult procedure to learn. Knowledge of proper handling of bacterial specimens is necessary. Cannot be performed in the field.

Ordering information

<ins>Vendor</ins>: IDEXX
1 IDEXX Drive
Westbrook, ME 04092
Phone: 1-800-321-0207
Fax: 207-856-0630
E-mail: water@idexx.com
Website: www.idexx.com/water

Equipment/Supplies Needed for *E. coli* Analysis

Item (Catalog Number)[1]	Quantity	Price*
Colilert reagent for 100mL sample (WP200)	200-pack	$1,020.00
120mL vessel with 100mL line, sodium thiosulfate & label (WV120ST-200)	200-pack	$90.00
97-well sterile Quanti-Tray/2000 trays (WQT-2K)	100-pack	$110.00
Quality control kit (E. coli, Klebsiela, Pseudomonas A). (WKT 1001)	n/a	$120.00
Colilert comparator predispensed in a Quanti-Tray/2000 (WQT2KC)	1	$6.00
Quanti-Tray Sealer (115V) with 51-well rubber insert (WQTS2X-115)	1	$3,500.00
6 watt UV lamp 110 volt (WL160)	1	$89.00
Incubator 120V, 30-65°C, 14"x14"x14" (WI300)	2	$389.00

[1] See the Enterococci table above for equipment that can be shared when conducting both analyses.

**The per-sample expendable cost (reagent, bottle, and tray) is about $6.65.*

Equipment/Supplies Needed for Enterococci Analysis

Item (Catalog Number)	Quantity	Price*
Enteroletert reagent for 100 mL samples (WENT200)	200-pack	$ 1,020.00
120 mL pre-sterilized vessel with 100 mL line, sodium thiosulfate & label (WV120ST-200)[1]	200-pack	$ 90.00
97-well sterile Quanti-Tray/2000 trays (WQT-2K) [1]	100-pack	$ 110.00
Quality control kit (E. coli, Klebsiela, Pseudomonas A). (WKT 1001)	n/a	$ 120.00
Quanti-Tray Sealer (115V) with 51-well rubber insert (WQTS2X-115) [1]	1	$ 3,500.00
6 watt UV lamp 110 volt (WL160) [2]		

F2.8 FLUORIDE (0 TO 2.00 MG/L F^-)

Equipment/Supplies Needed

- Hach bench top or portable spectrophotometer or colorimeter (see ordering information below)
- AccuVac Vial Adaptor (for older spectrophotometers)
- SPADNS Fluoride Reagent AccuVac Ampuls.

Procedure

Refer to Hach SPADNS Method 8029 which is adapted from Standard Methods for the Examination of Water and Wastewater. This procedure involves the reaction of fluoride with a red zirconium-dye solution. The fluoride combines with part of the zirconium to form a colorless complex, thus bleaching the red color in an amount proportional to the fluoride concentration.

Duration of Test for Each Sample

Each samples takes an average of 3 minutes to test.

Hazardous Reagents

The SPANDS reagent is a hazardous solution. The used AccuVacs should be placed back in the Styrofoam shipping container for storage and then disposed properly through a hazardous waste disposal company.

Ease of Analysis

The procedure is relatively easy and fast and can be performed in the field using a portable spectrophotometer or colorimeter. However, as for all tests, it is recommended that the analyses be conducted in a laboratory, or at least in a work room having good lighting and water.

Ordering information

Vendor: Hach Company
PO Box 389
Loveland, CO 80539-0389
Tel: 800-227-4224
Fax: 970-669-2932
Website: www.hach.com

Equipment/Supplies Needed for Fluoride Analysis

Item (Catalog Number)	Price
Fluoride Reagent (SPADNS) AccuVac Ampuls [1 set of 25 AccuVacs (2 needed per test)] (2506025)	$ 17.00
Adapter, AccuVac vial (needed for older spectrophotometers DR/2000 and DR/3000) (43784-00)	$ 5.40
DR/890 portable colorimeter programmed with 90 tests. Includes 2 sample cells, COD & TnT tube adapter, instrument, procedure manual and batteries. Portable instrument that can be used for many different analytes, but fewer than the following instruments. (48470000) [1]	$ 929.00
DR/2500 spectrophotometer includes 6 one-inch round sample cells, instrument and procedure manual, and DR/Check Absorbance Standards. Compact laboratory instrument having many capabilities. (5900000) [1]	$ 2,200.00

F2.9 pH

Equipment/Supplies Needed

- Cardy pocket-sized pH meter model B-213 made by Horiba
- pH standards that come with the meter.

Calibration

The meter should hold its calibration for an extended period, but it is best to check the calibration before each sample batch.

1. Press the ON/OFF button.
2. Place approximately 1 mL of the yellow pH 7.0 standard solution onto the sensor cell (be careful not to touch the sensor with the dropper or pipette, the cell is covered with a very thin and fragile glass cover slip).
3. Press the CAL button to display the black CAL mark in the upper right corner and 7.0.
4. Calibration is complete when the CAL mark disappears. Wash the sensor with tap or distilled water and dry with a tissue.
5. Press CAL again so that 4.01 and CAL are displayed to calibrate using the pink pH 4.01 buffer. Follow the same procedure as above.

Measurement

1. Place a drop of the sample water

F2.10 POTASSIUM

Equipment/Supplies Needed

- Cardy potassium compact meter by

Equipment/Supplies Needed for pH Analysis	
Item (Catalog Number)	**Price**
Cardy potassium compact meter and accessories (EW-05755-00)	$239.00
Replacement cardy potassium sensor cartridge (EW-05755-500)	$ 64.00
Replacement cardy potassium solution kit (EW-05755-60)	$ 33.00

Note: This procedure is rapid and inexpensive, however, it only has a detection limit of about 1 mg/L, and reads in increments of 1 mg/L. This level of precision is not typically a problem for moderately contaminated samples (when the results are most useful); however, it presents challenges when used for cleaner water.

F2.11 TOTAL HARDNESS (10 – 4000 mg/L as CaCO3)

Equipment/Supplies Needed

- Hach digital titrator
- Total hardness titration cartridge
- ManVer 2 hardness indicator
- Hardness 1 buffer solution.

Procedure

Refer to Hach Method 8213 for Hardness, Total (10-4000 mg/L as $CaCO_3$) digital titrator method using EDTA. This procedure involves buffering the sample first to pH 10.1, adding of the ManVer 2 Hardness Indicator, which forms a red complex with a portion of the calcium and magnesium in the sample, and then titrating with EDTA. The EDTA titrant reacts first with the free calcium and magnesium ions, then with those bound to the indicator, causing it to change to a blue color at the end point.

Duration of Test for Each Sample

Approximately 5 minutes.

Hazardous Reagents

The mixture of sample, buffer solution, hardness indicator, and EDTA must be stored properly in a labeled container until disposal by a hazardous waste disposal facility.

Ease of Analysis

This procedure is not recommended to be performed in the field. Produces hazardous chemicals.

Ordering information

Vendor: Hach Company
PO Box 389
Loveland, CO 80539-0389
Tel: 800-227-4224
Fax: 970-669-2932
Website: www.hach.com

Equipment/Supplies Needed for Total Hardness Analysis		
Item (Catalog Number)	**Quantity**	**Price***

F2.12 TURBIDITY

Equipment/Supplies Needed

- Benchtop or portable turbidimeter. The range of readings in NTU will depend upon the instrument.

Procedure

(This is a general procedure for turbidity. Follow your turbidimeter’s instructions):

1. First, the instrument must be calibrated using the standards supplied with the instrument. If calibration is satisfactory, continue with sample measurement.
2. Samples are normally stored under refrigeration. Before analyzing for turbidity, the samples mu000.uNn1(it-1.1(nrt be6.1(e -0.61()]TJT*0.0003 Tc-0.0005 Tw[(1brough beack t

w3 68lPur tue staNn1(m)pl insto a am

Appendix F3. METHODOLOGIES AND LAB TESTING OF TECHNIQUES TO MEASURE DETERGENTS

F3.2 FLUORESCENCE MONITORING USING THE GFL-1 FLUOROMETER

Introduction

Fluorescence is the property of the whiteners in detergents that cause treated fabrics to fluoresce in the presence of ultraviolet rays, giving laundered materials an impression of extra cleanliness. These are also referred to as bluing, brighteners or optical brighteners

Calibration

Before the instrument is used, it should be calibrated with a detergent solution. No general standard detergent solution is available, so a commercially available detergent is used to prepare standard solutions. For this research, a common commercial detergent, Procter & Gamble's *Tide* ™ was used. The purpose of calibrating the fluorometer is to set the instrument fluorescent signal levels to correspond to different concentrations of this commercial le's

[illegible] 9(h a de commT*0.0003 Tc-0.0013 Tw[(Beforesceo1(e)-0.]T

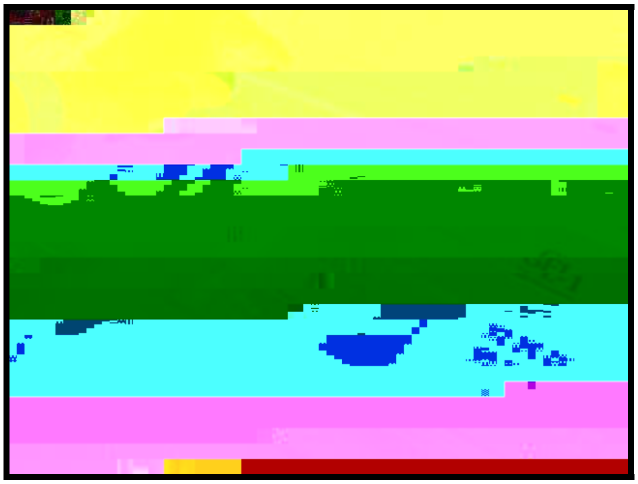
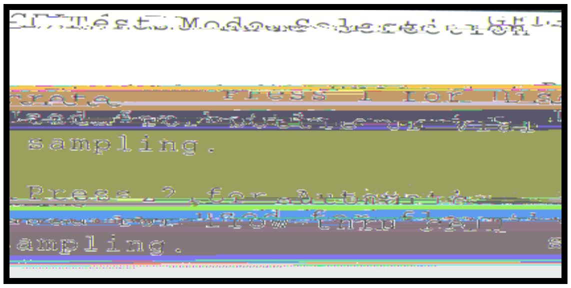


Figure F3.5: Placing Sample into Sample Chamber

Log File:

Cal File:

Figure F3.7: Discrete Sample Mode



Test Mode Selection

Press 1 for Dis

sampling.

Press 2 for Automatic

flow thru

sampling.

Figure F3.6: Test Mode Selection

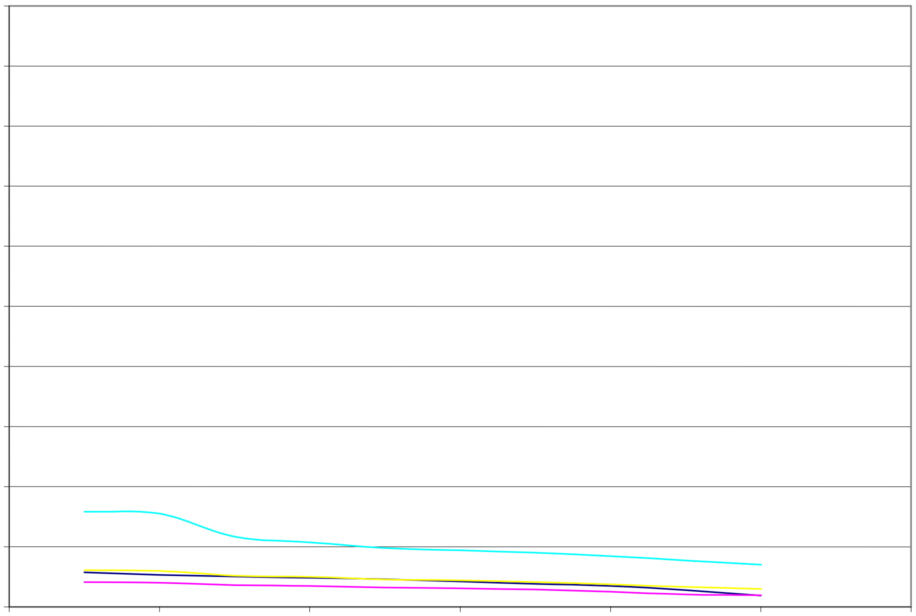
Press 1 for doing discrete bottle sampling. A new screen will appear (Figure F3.7).

With calibration complete, the instrument is ready to analyze the samples. To run a test, simply load a sample into the chamber and press ENTER. The unit will measure the sample and present the data a few seconds later. A busy message indicates that the test is in progress. Press ESC to return to the main menu.

Initial Tests using the Fluorometer

Initial tests were conducted after the first calibration to get an indication of the repeatability and drift of the results obtained from the new instrument. Five different concentrations of Tide detergent samples were made and tested for fluorescence after varying periods of time. The results of these tests are shown in Figure F3.8.

It is obvious that the fluorescence signal from Tide degrades with time and that the analyses should be evaluated within two hours. Other samples of commercial and household detergents were also evaluated and degradation of fluorescence with time was also identified. The largest changes occurred between about one and two hours after sample preparation. There was very little change after this initial two hour period. In the real world, the time between mixing of a laundry detergent with the washwater at the laundry, its discharge, and its analysis in the laboratory is at least two hours. Therefore, the fluorescence values used are those obtained after the signals have reached a relatively constant value. The results of the tests on certain commercial and household detergents are shown in Figure F3.9.



The results indicate that this method can be used as a presence/absence test for detergent concentrations between 0.2 and 1 mg/L (as Tide) and to estimate concentrations above 1 mg/L. The method is simple and does not require specialized equipment.

An advantage of this method is that the equipment is easily available and inexpensive. The disadvantages are the variability in readings due to changes in temperature and characteristics of the detergents.

Figure F3.10 shows the results from concentrations between 10 and 50 mg/L. For readings above 10 mg/L, if the level of detergent increases the height of the foam also increases in a parabolic shape. It was also observed that the repeatability of the results decrease at high levels.

For levels of detergent lower than 10 mg/L, there is not an important change in the reading. The minimum reading that can be obtained from the burette is 0.05 mL. For samples in this range the reading is close to the precision of the instrument. Figure F3.11 shows the results from concentrations between 0 and 5 mg/L.

Readings below 1.0 mg/L create a circle of bubbles around the wall of the pipette. This circle was not present when distilled water was used. This procedure can be used as a presence/absence test. The circle was observed for concentration of detergent higher than 0.2 mg/L.

Conclusions

The new method is an inexpensive, safe and moderately accurate method to estimate the presence of detergents in concentrations above 0.2 mg/L. For detergent concentrations above 10 mg/L, the method can be used to quantify the concentrations. These higher concentrations have been observed in sewage, industrial discharges, laundries and car wash areas.

Table F3.11: Foam Readings Over Time

Concentration (mg/L, as Tide)	Foam Height after 10 sec. (mL)	Foam Height after 1 min. (mL)
0	0	0
0.1	0	0
0.2	T	T
0.3	T	T
0.4	T	T
0.5	T	T
0.7	T	T
1	0.05, 0.05, 0.05, 0.05	0.05, 0.05, 0.05, 0.05
2	0.1, 0.1, 0.1, 0.1	0.1, 0.1, 0.1, 0.1
3	0.1, 0.1, 0.15, 0.15	0.1, 0.1, 0.15, 0.15
5	0.15, 0.15, 0.15, 0.15	0.15, 0.15, 0.15, 0.15
10	0.2, 0.2, 0.2, 0.2	20 0.35, 0.4, 0.4, 0.4 0.8, 0.6, 0.6, 2 00.13, 1.4
50	2.6, 2.6, 3.0, 2.8	3.8, 3 0, 3.7, 3.6

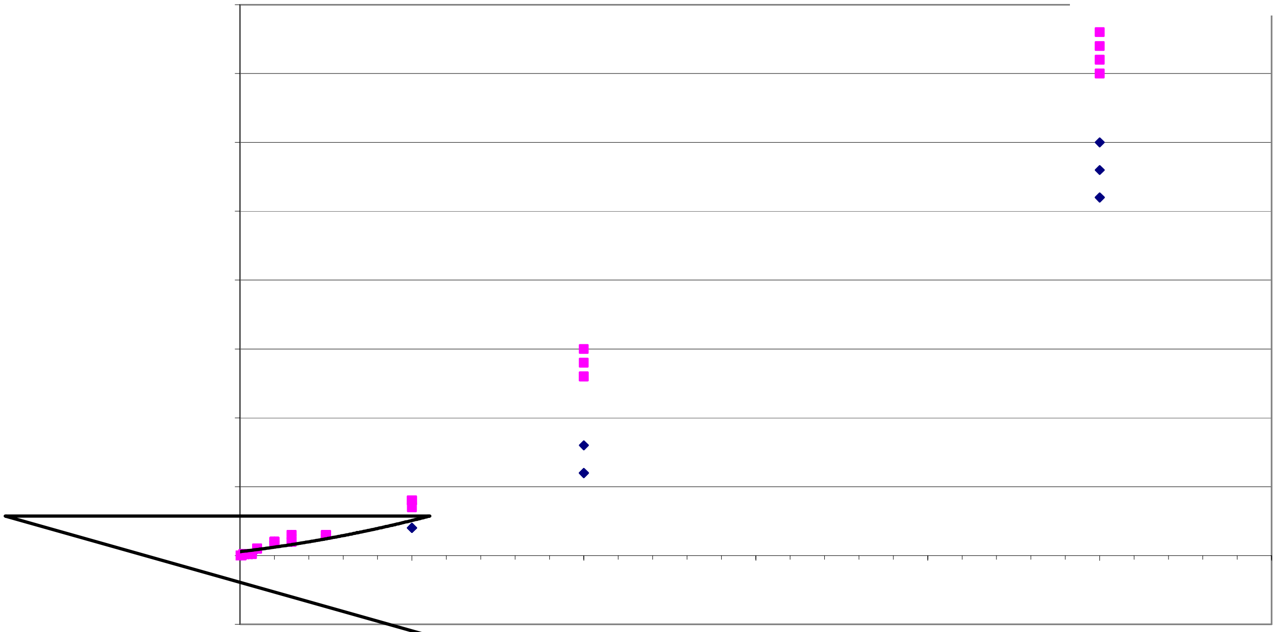


Figure F3.10: Correlation Between Concentration and Foam Height at Higher Concentrations Concentration Vs Foam Heighty

APPENDIX F4: LAB TESTING OF "OPTICAL BRIGHTENER MONITORING" TO FIND INTERMITTENT DISCHARGES

Introduction

Fabric brighteners are fluorescent dyes added to soaps and detergents. These are used to produce a brightening effect after laundering. They absorb the UV rays of the sunlight and then fluoresce as a bright blue.

Optical Brightener Monitoring (OBM) is a new method for detecting fluorescent materials in water samples. It is based on a method used to measure the presence of strongly fluorescent tracer dyes.

Briefly, cotton pads that are free of fabric brighteners are used for checking the presence of optical brighteners in water samples. Cotton pads are soaked in the water sample and then dried in a darkened room. The pads are then viewed with ultraviolet (UV) light to check for the presence of fluorescence. This is an inexpensive, but much less sensitive, method for the detection of fluorescence compared to fluorometers.

Homemade OBM traps are inexpensive and easy to make. Table F4.1 lists the average

Test Procedure

Step One:
Care should be taken so that samples are handled properly with no cross

Figure F4.2/4east a onee houaTv

Method Modifications

While reviewing the prior methods for the OBM for inappropriate discharge detection, the following issues were brought up:

a) Do the pads need to be left in the field for extended periods and how long should the pads be exposed to the sample water?

b) Are there any detrimental effects of direct exposure to sunlight while drying the cotton pads?

c) What is the sensitivity of the OBM compared to the other tests used to detect washwaters and sanitary wastewaters?

The above points are discussed in the following paragraphs.

Leaving the cotton pad and the sampling device at the sampling location

If there is continuous flow at an outfall, there is no need to keep the pads at the outfall for extended periods. If grab samples are collected from the flowing outfalls for later chemical tests, a separate sample bottle can be conveniently collected for optical brightener tests. During our analyses, the cotton pads were immersed in the sample bottles at the time of sample collection. This sampling modification greatly reduced the time and effort needed to conduct the tests. Our initial tests indicated that the high sediment loads associated with the outfall discharges would hinder the ability to measure the fluorescence due to coating the fabrics with silt. If the pads were placed in the OBM sample bottles when the water was collected, the time required to bring the samples to the laboratory was thought to be sufficient to affect the pads. Tests were conducted in the laboratory to determine the time needed to affect the pads. The standard procedure used at least a one hour exposure period.

Direct exposure to sunlight while drying the cotton pads.

There was a concern related to the degradation of fabric fluorescence in the presence of sunlight, especially after the fluorometer tests indicated significant decreases in water sample fluorescence during the first hour or two after detergent mixing. In order to test this concern, two samples were prepared with the same concentration of detergents. Two cotton pads were immersed in each of the bottles. One was dried under the direct exposure of sunlight, while the other one was dried in a dark room. After 24 hours, both sets of pads gave the same fluorescence under the ultraviolet light. Therefore, it was concluded that direct sunlight exposure to the dried cotton pads did not affect the test results.

Other sampling and laboratory practices that were important included using gloves while handling the pads, and testing the cotton pads for fluorescence under the UV lamp before their use.

Laboratory Verification using Standard Samples and Field Use in Cribbs Mill Creek

The basic OBM method is a presence/absence test, with unknown sensitivity. In order to make this test more useful, additional tests were conducted. The initial test used different Tide detergent standards. Tide detergent samples were made with concentrations of 0.5 mg/L, 5 mg/L, 10 mg/L, 20 mg/L, 30 mg/L 50 mg/L, 100 mg/L, and 500 mg/L. Samples from each dried test pad were attached onto a card, as shown in Figure F4.4.

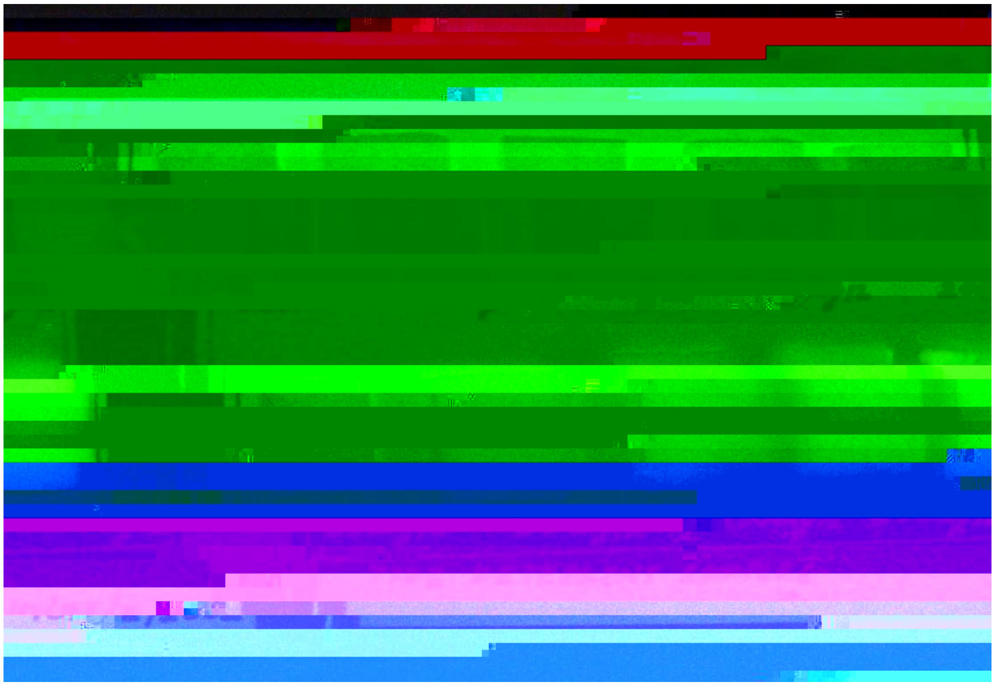


Figure F4.4: Standard Tide OBM Pads

As can be seen in Figure F3.4, concentrations below 35 mg/L all look identical. The 50 mg/L Tide solution (the first one with an obvious fluorescence response) is representative of a full-strength washwater as typically used in household laundry. Thus, it may be concluded that the OBM method may not be useful for samples having anything less than full-strength washwaters.

The maximum fluorescence concentration obtained from the Cribbs Mill Creek samples was 17mg/L (as Tide), and no positive responses for fluorescence using the OBM method were found.

Conclusion

This test was originally designed to identify faulty septic systems and storm drainage systems using fluorescent dyes. The fluorescent dyes (Fluorescence and Rhodamine FWT) used in these types of tests are very strong dyes and are used in moderate concentrations. They are therefore much easier to be detected by the cotton pads and the OBM method than the fabric brighteners in washwaters. OBM is a quick, easy, and inexpensive method, but can only reliably detect undiluted washwaters, and likely will miss the more common diluted washwaters found as inappropriate discharges. Other simple methods exist that are more sensitive, although the OBM method may be most suitable if intermittent discharges of undiluted washwaters are expected.

Appendix F5. IN-HOUSE ANALYTICAL CONSIDERATIONS FOR INDICATOR

Introduction

Program managers need to understand the basic analytical options and safety considerations, for each analytical method used to measure indicator parameters. This understanding helps program managers choose what indicator parameters to collect and where they should be analyzed. This section provides a summary of the basics.

Table F5.1 summarizes the recommended analysis method associated with each indicator parameter. An extended description of each analysis method is provided below.

Colorimetric – Colorimetric methods utilize specialized instruments such as a colorimeter or a spectrophotometer (Figure F5.1). The two instruments are similar and quantify parameter concentrations by adding reagents to the sample and passing through a defined spectrum of light. In general, spectrophotometers can analyze a much broader range of parameters than colorimeters.

Table F5.1: Analytical Considerations for Illicit Discharge Indicator Parameters

Indicator Parameter	

Figure F5.1: Spectrophotometer

Color Comparator – This analysis method is a less quantitative version of the colorimetric method. Samples are prepared by adding reagents, and assessing the color in comparison to a color cube (see Figure F5.2) or color disk that assigns a concentration for different color shades.

Figure F5.2: HACH Color Cube Comparator

Probes – These methods use a probe to pass an electrical current through the sample for specific light wavelength (for most indicators) or measure the scatter of light (for turbidity). While results are immediate, lab analysts need to frequently calibrate the probe using standard solutions to assure accurate data.

Titration – Titration techniques measure the concentration of indicator parameters by determining the amount of a reagent needed to produce a specific reaction in the sample, which is often indicated by a color change. Lab analysts carefully record the amount of reagent added to the sample using a "burette," which is a graduated cylinder with a valve-controlled opening at the bottom. An alternative and more precise technique is a digital titrator. Both methods rely on equations or lookup tables that relate to the amount of reagent added to the estimated concentration of the indicator parameter.

IDEXX Techniques: Colilert or Colisure - These proprietary methods are used to measure *E. coli*, total coliform and Enterococci bacteria. Samples are sealed along with a reagent in a specialized tray that is then placed into an incubator for 24 hours. The analyst then measures the number of cells in the tray that have changed color or shine under a fluorescent bulb, which is used to indicate the amount of bacteria in the sample (Figure F5.3). The IDEXX method uses a standard chart to relate the number of cells that have a positive reaction to the presence of bacteria. The IDEXX method is fairly simple and safe, but requires fairly expensive equipment.

Safety and Waste Management Considerations

Each analysis method has special safety and waste disposal considerations, which are outlined in Table F5.2.

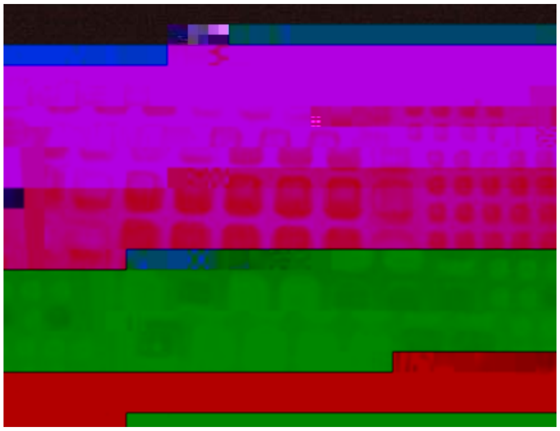


Figure F5.3: IDEXX Results

APPENDIX G

APPENDIX G

Developing a Consistent Sample Collection Protocol

A good field sampling protocol incorporates eight basic elements:

1. Where to collect samples
2. When to collect samples
3. Sample bottle preparation

pollutants. Some considerations for sample collection include:

- Wear surgical gloves (unpowdered nytrile gloves are recommended to limit

Table G.1: Sample Preservation and Storage Requirements for Typical Outfall Monitoring Parameters

(Primary Source: APHA, 1998)

Parameter	Preservation[3]	

Table G.2: Nine Elements of a Chain of Custody

Element of Chain of Custody	Description
1. Sample Labels	

7. Quality Assurance Measures During Sample Collection

To ensure sampling results are accurate, it is important to institute quality assurance measures as part of the sampling protocol. Quality assurance samples serve as a check against biases introduced during sample collection, or within the laboratory. Quality assurance samples also assess the accuracy of the analysis method and its consistency for samples collected at the same site. The sampling protocol should define a minimum fraction of samples that will be used for quality assurance purposes (typically about 5% - 10% of all samples collected). Examples of quality assurance samples include field blanks, duplicate samples, split samples and spiked samples, which are described below:

Field Blanks – Field blanks are deionized water samples prepared in the field at the time of sample collection. If the lab results for field blanks have non-zero values, it indicates that impurities were introduced to the sample during collection or lab analysis. The distilled deionized water should be placed in whatever is used to collect samples (e.g., sample scoop, dipper, plastic milk bottle) and then poured in the sample bottle, just as if it had been scooped or dipped as a real sample.

Duplicate (Replicate) Samples – This quality assurance technique relies on the collection of two or more samples from the same location and flow source during the same field visit. A discrepancy between the two sample measurements indicates a lack of precision or repeatability introduced during sample collection or lab analysis.

Field Spikes – A field spike is a sample to which a known concentration of an indicator parameter is added (e.g., an ammonia concentration of 1.0 mg/L). Any difference between the known concentration and the final laboratory measurement reveals errors introduced during sampling and laboratory analysis.

Split Samples – Splits consist of a single field sample that is divided into two separate sub-samples for subsequent laboratory analysis. Typically, split samples are submitted to different laboratories, or analyzed by different analysts to determine the precision of laboratory results. Alternatively, split samples can be analyzed at a single laboratory without knowledge of the sample origin (referred to as a "blind sample"). Any discrepancy between the two sub-samples suggests a lack of precision or repeatability introduced during sample collection or lab analysis.

8. Safety Considerations

Whenever sampling is done there are safety considerations that require planning. This is even more important when sampling is being conducted in urban stream environments where there is potential for contact with contaminated water, sharp debris and objects, and threatening individuals (both animals and humans). Field crews should be comprised of at least two individuals, each equipped with proper foot (e.g., sturdy boots or waders) and hand wear (latex gloves). Key equipment for crews to carry include cell phones, a list of contact and emergency numbers, a gps unit, and a first aid kit. Private properties should not be accessed unless proper notification has been provided, preferably in advance. Lastly, program managers may want to consider requiring/recommending field crews to be vaccinated against Hepatitis B, particularly if the crews will be accessing waters known to be contaminated with illicit sewage discharges.

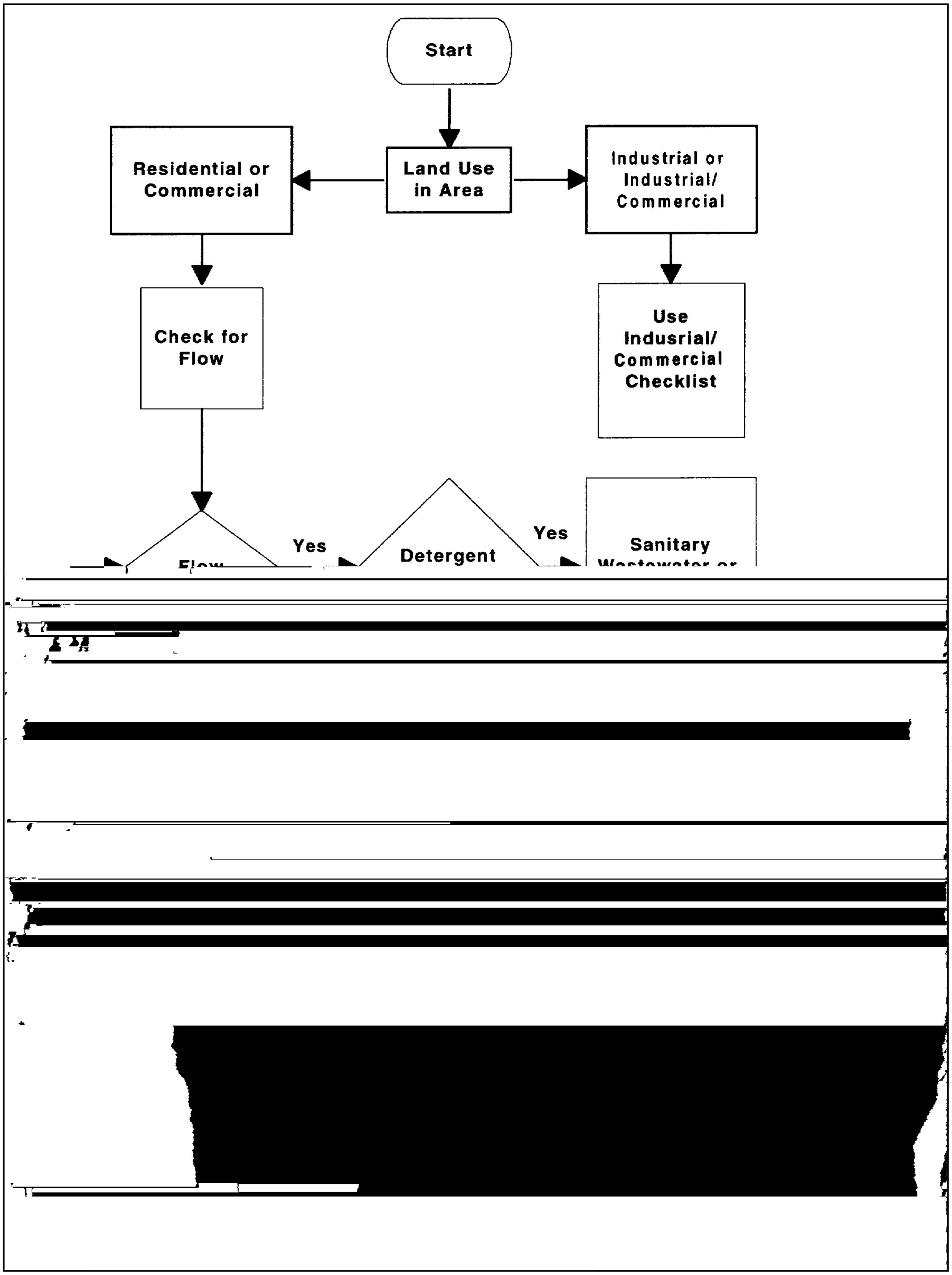
References

American Public Health Association (APHA).1998. *Standard Methods for the Examination of Water and Wastewater – 20th Edition*. Washington, D.C.

APPENDIX H

TWO ALTERNATIVE FLOW CHARTS

Figure H.2 Original Flow Chart Derived from Data in Birmingham
(Pitt and Lalor, 1993)



APPENDIX I

USER'S GUIDE FOR THE CHEMICAL MASS BALANCE MODEL
VERSION 1.0

(Adapted from Karri, 2004)

Overview of the Model

The Chemical Mass Balance Model (CMBM) estimates the most likely source components that contribute to outfall flows during dry weather. In order to use the model, the user 6008 Tm(ates thc92.579557 In o648 122.28496 680.7y7p. 10.02 2ETBT7TT o648 122.28496 680

Library File Format

This model recognizes the source file for evaluation, only when it

Example Problems

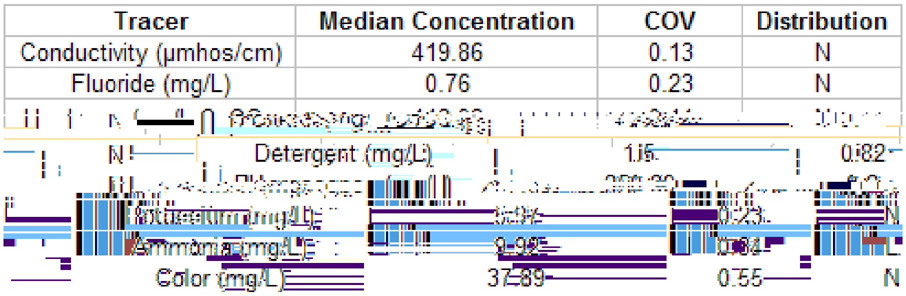
Example 1

This first example illustrates a verification procedure that is used to ensure the model is functioning as expected. It assumes the analysis of an undiluted flow.

Consider an outfall, which has the same data for the tracer parameters as were observed at the sewage treatment plant (which is the same as the library data for sewage wastewater). This means that the model must predict the most likely source component to be sewage and with a predicted fraction of flow for sewage close to one.

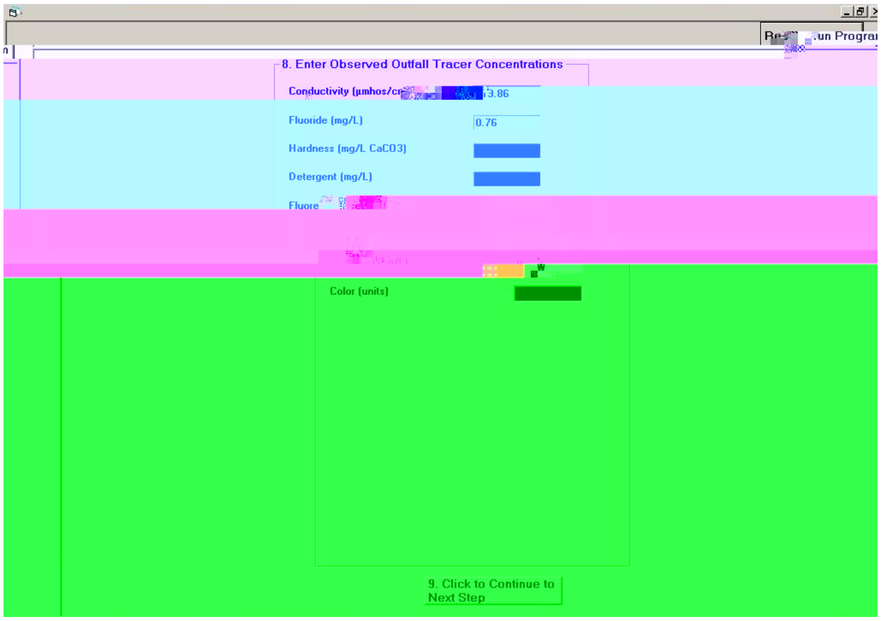
The library file used here is the Birmingham library file 'Library_BHM.xls' (which is included with the program). Let the number of Monte Carlo simulations considered be 1000, and the number of sources selected for evaluation be 4 (sewage wastewater, tap water, spring water, and landscape irrigation runoff). The tracers selected are conductivity, fluoride, potassium and ammonia. Figure I.5 shows these corresponding entries, while Figure I.6 shows the Excel spreadsheet for the library file used.

Figure I.7 shows the entries made in the second form. It should be noted that the values for the tracers entered are the same as those in the librefilyj12 0 0 12 5214520.5057591 29529o4lib



Tracer	Median Concentration	COV	Distribution
Conductivity (µmhos/cm)	419.86	0.13	N
Fluoride (mg/L)	0.76	0.23	N
[illegible]	[illegible]	[illegible]	[illegible]
Detergent (mg/L)	1.15	0.82	[illegible]
[illegible]	[illegible]	[illegible]	[illegible]
Potassium (mg/L)	5.97	0.23	N
Ammonia (mg/L)	9.92	0.34	L
Color (mg/L)	37.89	0.55	N

Figure I.6: Library File Excel Sheet (Sewage Wastewater)



8. Enter Observed Outfall Tracer Concentrations

Conductivity (µmhos/cm) [illegible]3.86

Fluoride (mg/L) 0.76

Hardness (mg/L CaCO3)

Detergent (mg/L)

Fluore[illegible]

Color (units)

Figure I.7: Form 2 (Input)

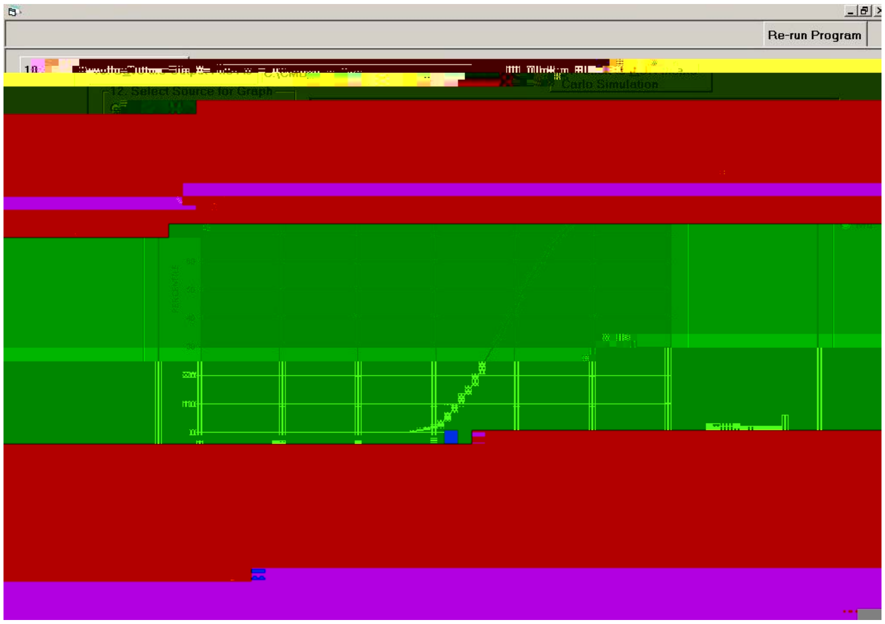


Figure I.8: Form 3 (Output for Example 1)

Example 2
In this example, eight possible source types and eight tracer parameters are selected, based on sample data from outfall # 20 in Birmingham, AL, collected on March 3, 1993.

The library file used in this example is also

Figure I.10: Form 2 (Input for Example 2)

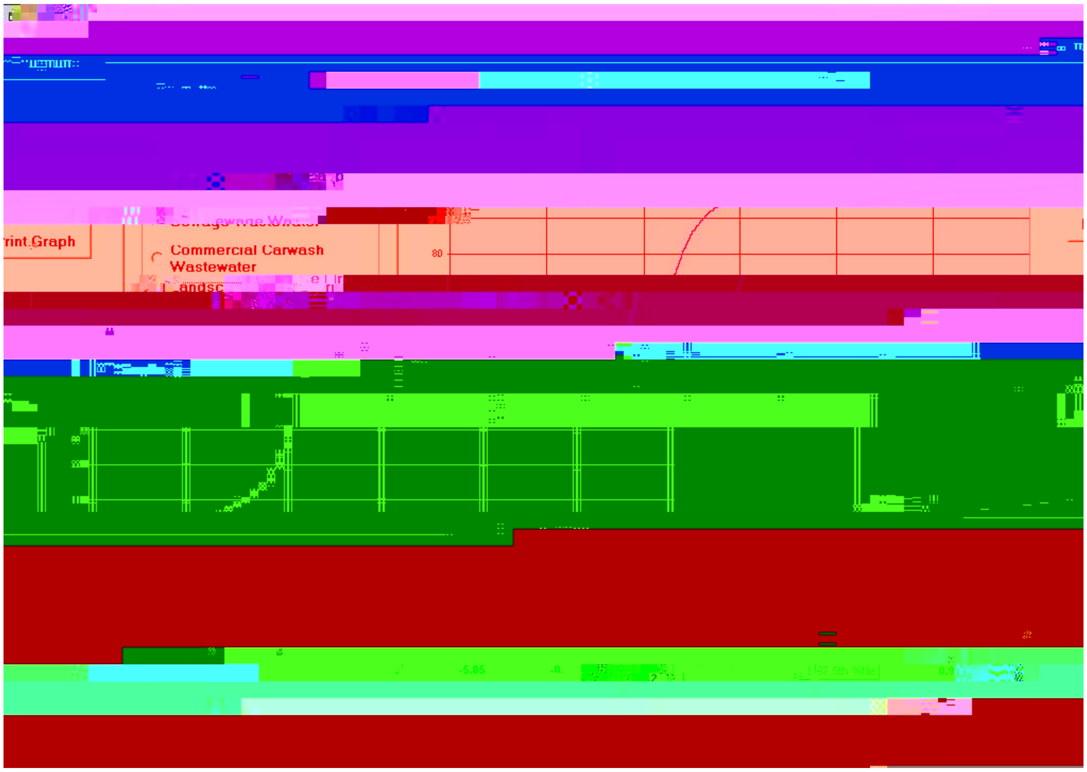


Figure I.11: Form 3 (Output for Example 2)

APPENDIX J

USING THE CHEMICAL LIBRARY TO DETERMINE THE UTILITY OF BORON AS AN INDICATOR OF ILLICIT DISCHARGES

Introduction

In this example, library data from several flow types are analyzed to determine a good cut-off point to use boron as an indicator of illicit discharges. Both the data and the selected concentrations are derived from research in Tuscaloosa, Alabama (Pitt, 2004). Investigators examined the data from their chemical flow library both graphically and then in detail to select a concentration.

Step 1: Visually Analyze Data Using Box Plots

After collecting data from a select group of flow types, researchers assembled the data into box plots (see Plots 1 and 2). These plots help quickly identify the range of data. The “box” portion of the plot shows the first quartile, median, and third quartile for the data, and the individual data points show the data above and below this range.

A first look at the data shows that sewage, laundry, and wash water sources all have a higher concentration than the non-illicit flows: irrigation, tap water, and spring water. A closer look, using the log plot (i.e., the log of each concentration), shows some overlap between irrigation water and two of the illicit flow types: laundry and car wash. Although this overlap means that there will be some “false negatives” or “false positives” using this parameter, investigators select a concentration that is lower than the lowest concentration in laundry. This value appears to be somewhere between $10^{-0.5}$ (or 0.3 mg/L) and 10^{0} (or 1.0 mg/L).

Carwash Sewage

Boron Concentration (mg/L) For Six Flow Types
(Concentrations >0.35 mg/L indicate illicit discharges)

Tap Water	Spring Water	Irrigation	Laundry	Car Wash	Sewage
0.04	0.04	0.13	0.36	0.09	0.78
0.1	0.09	0.14	0.53	0.28	0.93
0.11	0.09	0.14	0.58	0.37	0.97
0.12	0.14	0.2	0.67	0.48	0.98
0.14	0.15		0.7	0.5	1.01

References

Pitt, R. 2004. *Methods for Detection of Inappropriate Discharge to Storm Drain Systems*. IDDE Project Support Material.

APPENDIX K

SPECIFIC CONSIDERATIONS FOR INDUSTRIAL SOURCES OF INAPPROPRIATE POLLUTANT ENTRIES TO THE STORM DRAINAGE SYSTEM

(Adapted from Pitt, 2001)

Industrial Site Surveys

Additional pollutants associated with local commercial and industrial activities need to be monitored during outfall screening activities if these activities exist in the watersheds of interest. This monitoring will assist in identifying the classes of commercial or industrial activities responsible for the contamination. The first step in this process is to identify which industrial and commercial activities may contribute non-storm water discharges to the drainage system. The review of industrial user surveys or reports that are available needs to be done initially. It may be necessary to also send a questionnaire to industries in the watershed that are draining to the storm drainage system to identify the specific activities that may affect runoff quality and dry weather discharges. Site inspections will still be required because questionnaires may not be returned or may give incorrect details (either deliberately or unknowingly).or particular industry. It should be noted that a combination of just a few of these characteristics, or perhaps all of them, might occur at an outfall affected by a potential source. In addition, outfalls are likely to be affected by several sources simultaneously, further confusing the situation. Again, a complete watershed analysis describing the industrial and commercial facilities operating in each outfall watershed will be of great assistance in identifying which industries may be contributing harmful dry weather discharges to the storm system.

City: ___________________________ Industry Name: ______________________________________

Table K.1: Chemical and Physical Properties of Industrial Non-Storm Water Discharges

Industrial Categories Major Classifications SIC Group Numbers	Odor	Color	Turbidity	Floatables	Debris and Stains	Structural Damage	Vegetation	pH	Total Dissolved Solids

Table K.1: Chemical and Physical Properties of Industrial Non-Storm Water Discharges

Industrial Categories Major Classifications SIC Group Numbers	Odor	Color	Turbidity	Floatables	Debris and Stains	Structural Damage	Vegetation	pH	Total Dissolved Solids
30 Rubber & Miscellaneous Plastic Products	Rotten Eggs, Chlorine, Peroxide	Brown to Black	Mod.	Shredded Rubber Pieces of Fabric or Metal	Gray to Black	Low	Inhibited	Wide Range	High
Transportation & Construction									
15 Building Construction	Various	Brown to Black	High	Oils, Grease, Fuels	Gray to Black	Low	Normal	Normal	High
16 Heavy Construction	Various	Brown to Black	High	Oils, Grease, Fuels, Diluted Asphalt or Cement	Gray to Black	Low	Normal	Normal	High
Retail									
52 Building Materials, Hardware, Garden Supply, and Mobil Home Dealers	NA	Brown to Black	Low	Some Seeds, Plant Parts, Dirt, Sawdust, or Oil	Light Brown	Low	Normal	Normal	Low
53 Gen. Merchandise Stores	NA	NA	NA	NA	NA	Low	Normal	Normal	Low
54 Food Stores	Spoiled Produce, Rancid, Sour	Various	Low	Fragments of Food, Decaying Produce	Light Brown	Low	Flourish	Normal	Low
55 Automotive Dealers & Gasoline Service Stations	Oil or Gasoline	Brown to Black	Mod.	Oil or Gasoline	Brown	Low	Inhibited	Normal	Low
56 Apparel & Accessory Stores	NA	NA	Low	NA	NA	Low	Normal	Normal	Low
57 Home Furniture, Furnishings, & Equip. Stores	NA	NA	Low	NA	NA	Low	Normal	Normal	Low
58 Eating & Drinking Places	Spoiled Foods Oil & Grease	Brown to Black	Low	Spoiled or Leftover Foods	Brown	Low	Normal	Normal	Low
Coal Steam Electric Power	NA	Brown to Black	High	Coal Dust	Black Emorphous Powder	Low	Normal	Slightly Acidic	Low
Nuclear Steam Electric Power	NA	Light Brown	Low	Oils, Lubricants	Light Brown	Low	Normal	Normal	Low

Other Chemicals Indicative of Manufacturing Industrial Activities

Table K.2 lists the various chemicals that may be associated with a variety of different industrial activities. It may be possible to examine non-storm water outfall flow for specific chemicals, such as shown on this list to identify which specific manufacturing industrial activities may be contributing the flows.

Example Problems for Locating an Industrial Source

Locating An Industrial Source

Hypothetical examples have been created to demonstrate how dry weather discharges can be characterized so that their likely industrial sources can be identified. These examples show how observations of outfall conditions and simple chemical analyses, combined with a basic knowledge of wastewater characteristics of industrial and commercial operations located in the drainage area can be used to identify the possible pollutant sources. The initial activities include pollutant analyses of outfalls being investigated. This requires the characterization on the non-storm water flows, the identification of the likely industries responsible for the observed discharges, and finally, locating the possible specific sources in the watershed.

The industries identified in a hypothetical storm water drainage area (from the watershed analysis) included a vegetable cannery, general food store, fast food restaurant, cheese factory, used car dealer, cardboard box producer, and a wood treatment company. The methods used to determine the most likely industrial source of the dry weather discharges are considered

deaterstr-upn1.T*3(p)0.outtdustrial and 1.145 TD0.000

Table K.2: Significant Chemicals in Industrial Wastewaters

Chemical	Industry
Acetic acid	Acetate rayon, pickle and beetroot manufacture
Alkalies	Cotton and straw kiering, cotton manufacture, mercerizing, wool scouring, laundries
Ammonia	Gas and coke manufacture, chemical manufacture
Arsenic	Sheep-dipping, fell mongering
Chlorine	Laundries, paper mills, textile bleaching
Chromium	Plating, chrome tanning, aluminum anodizing
Cadmium	Plating
Citric acid	Soft drinks and citrus fruit processing
Copper	Plating, pickling, rayon manufacture
Cyanides	Plating, metal cleaning, case-hardening, gas manufacture
Fats, oils	Wool scouring, laundries, textiles, oil refineries
Fluorides	Gas and coke manufacture, chemical manufacture, fertilizer plants, transistor manufacture, metal refining, ceramic plants, glass etching
Formalin	Manufacture of synthetic resins and penicillin
Hydrocarbons	Petrochemical and rubber factories
Hydrogen peroxide	Textile bleaching, rocket motor testing
Lead	Battery manufacture, lead mining, paint manufacture, gasoline, manufacture
Mercaptans	Oil refining, pulp mills
Mineral acids	Chemical manufacture, mines, Fe and Cu pickling, brewing, textiles, photo-engraving, battery manufacture
Nickel	Plating
Nitro compounds	Explosives and chemical works
Organic acids	Distilleries and fermentation plants
Phenols	Gas and coke manufacture; synthetic resin manufacture; textiles; tanneries; tar, chemical, and dye manufacture; sheep-dipping
Silver	Plating, photography
Starch	Food, textile, wallpaper manufacture
Sugars	Dairies, foods, sugar refining, preserves, wood process
Sulfides	Textiles, tanneries, gas manufacture, rayon manufacture

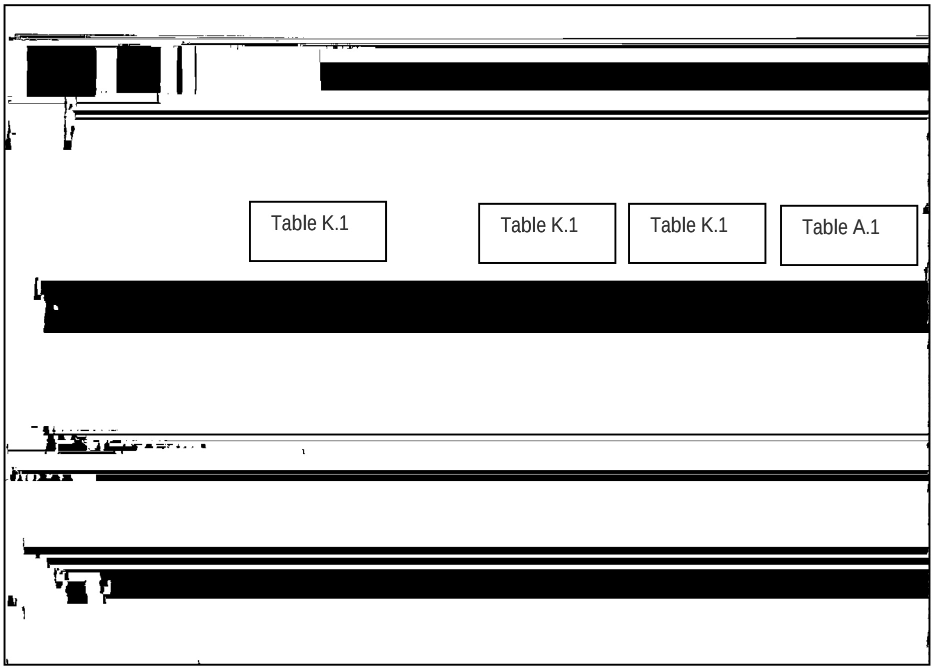


Figure K.3: Flowsheet for Case Example 2

Case Example 3

The results of the test measurements for the final situation found a normal pH (6) and low total dissolved solids (about 500 mg/L). Signs of contaminated discharges were found at the outfall only during and immediately following rainfalls. Other outfall properties observed included an odor of oil, deep brown to black color, a floating oil film, no structural damage, and inhibited plant growth (see Figure K.4).

According to Table K.1, the fast food restaurant and the used car dealer are the only two industrial sources in this hypothetical drainage area with a high potential for causing oily discharges. Their respective SIC categories are “Eating and Drinking Places” (SIC# 58) and “Automotive Dealers” (SIC# 55). Comparison of the properties shown in Table K.1 indicates inhibited vegetation only for the second category. Thus, the most likely source of the discharge is the used car dealer.

Furthermore, the source of contamination must likely be indirect, since the discharge occurs only during wet weather. Reference to Table A.1 (see Appendix A) under the category of “Car Dealers,” indicates a medium potential for indirect contamination. This fact, plus the knowledge that most used cars are displayed outdoors, makes it clear that surface runoff is probably carrying spilled automotive oil into the storm drain during rains.

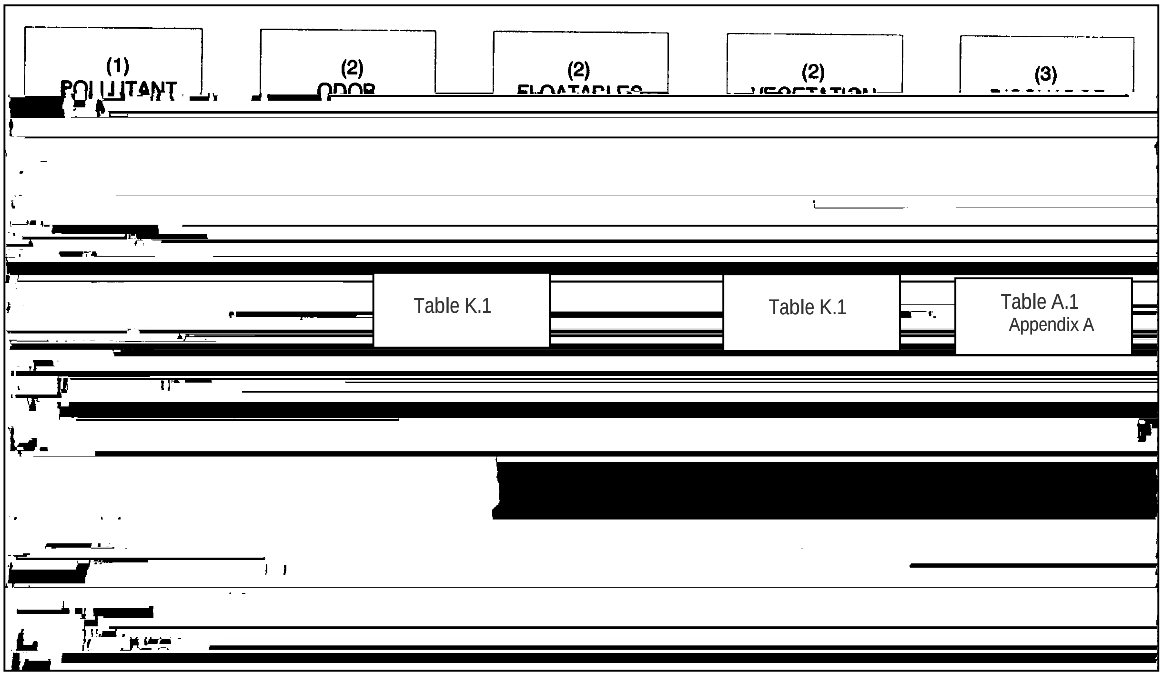


Figure K.4: Flowsheet for Case Example 3

References

Bannerman, R. 2003. Personal communication with Dr. Robert Pitt, University of Alabama.

Klein, L. 1962. "River Pollution 2: Causes and Effects." in D. Todd. 1979. The Water Encyclopedia. Water Information Center. Port Washington, N.Y.

Pitt, R. 2001. *Methods for Detection of Inappropriate Discharges to Storm Drainage Systems: Background Literature and Summary of Findings*. IDDE Project Support Material.